

THE RODENT UTEROTROPHIC ASSAY : CRITICAL PROTOCOL FEATURES,
STUDIES WITH NONYLPHENOLS, COMPARISON WITH A YEAST
ESTROGENICITY ASSAY

by

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SUMMARY

The major protocol features of the immature rat uterotrophic assay have been evaluated using a range of reference chemicals. The protocol variables considered include the selection of the test species and route of chemical administration, the age of the test animals, the maintenance diet used and the specificity of the assay for estrogens. It is concluded that three daily oral doses of estradiol, followed by determination of absolute uterine weights on the fourth day, provides a sensitive and toxicologically relevant *in vivo* estrogenicity assay. Rats are favoured over mice for reasons of toxicological practice, but the choice of test species is probably not a critical protocol variable, as evidenced by the similar sensitivity of rats and mice to the uterotrophic activity of methoxychlor. Vaginal opening is a useful extension of the uterotrophic assay for the detection of oestrogenic chemicals to reduce or abolish the uterotrophic response of estradiol is suggested to provide a useful extension of the uterotrophic assay for the detection of oestrogenic chemicals.

The results of a series of studies on the environmental estrogen nonylphenol (NP), and its hydrolysis product, the aliphatic side chain, are discussed. The hydrolysis product is shown to be active in the uterotrophic assay, and is suggested to represent the 'parent' estrogen of NP. Benzoylation of estradiol is shown to enhance its uterotrophic activity, in contrast to the enhancing effect of benzoylation on estradiol.

Selected chemicals shown to be active in the immature rat uterotrophic assay were also evaluated in an *in vitro* yeast human estrogen receptor transactivation assay. Most of the chemicals gave similar qualitative responses to those seen in the uterotrophic assay, and the detection of the estrogen methoxychlor by the yeast assay evidenced a degree of intrinsic metabolic competence. However, the assay had a reduced ability (compared to rodents) to hydrolyse the benzoate ester of estradiol, and the estrogenic benzoate derivative of NP was not active in the yeast assay. These last results indicate that current metabolic deficiencies of *in vitro* estrogenicity assays will limit the value of negative data for the immediate future.

The results described illustrate the intrinsic complexity of evaluating chemicals for estrogenic activities and confirm the need for rigorous attention to experimental design and criteria for assessing estrogenic activity.

INTRODUCTION

There is concern that some natural and synthetic chemicals have the potential to disturb normal sexual differentiation/development in animals and humans (Colborn and Clement, 1992; McLachlan, 1993; McLachlan and Korach, 1995). The ability of some chemicals to elicit an estrogenic response in intact organisms forms one of the primary mechanisms by which such adverse effects could be produced (McLachlan and Korach, 1995). One of the most extensively used assays for estrogenic activity is the rodent uterotrophic assay in which the ability of chemicals to stimulate uterine growth is monitored (Reel *et al.*, 1996; O'Connor *et al.*, 1996). There are a variety of protocols described for this assay, including the use of rats or mice; immature, hypophysectomized or ovariectomized animals, and the use of oral or subcutaneous routes of chemical administration. Given that most toxicity studies are currently conducted in rats, using the oral route of administration, we decided to evaluate the utility of the immature rat uterotrophic assay, using the oral route of chemical administration, as an *in vivo* assay for estrogenic activity. An associated objective was to discern the critical features of the assay protocol.

Having confirmed the practicality of the immature rat uterotrophic assay it was used to study a range of reference hormonally active chemicals and to elucidate the chemical basis for the estrogenic activity reported for branched chain r / p q p { n r j g p q n 0 " " H k p c n n { . " relationship between estrogenic activities observed *in vitro*, and those observed *in vivo*, some of the chemicals found positive in the uterotrophic assay were evaluated in the yeast human estrogen receptor (hER) transactivation assay described by Routledge and Sumpter (1996).

MATERIALS AND METHODS

Chemicals

Estradiol (E₂; for chemical structures, see Scheme 1), ethinyl estradiol (EE), cyproterone acetate (CPA), methoxychlor (MC; 98%), corn oil and arachis oil were obtained from Sigma (Poole, Dorset, UK). Estradiol benzoate (E₂B) was obtained from Intervet (Cambridge, UK) as a solution in arachis oil (peanut oil). Coumestrol was obtained from Apin Chemicals (Abingdon, Oxfordshire, UK) and ICI 182,780 was a gift from Dr A Wakeling, Zeneca Pharmaceuticals (Alderley Park, Macclesfield . " WM+ 0 " r / P q s obtained from p w q n " y c u q w t e g u < " H n w m c " E j g o k m D o r s e t , U K ; 8 5 % p a r k i s o n e r ' s : [N P (R)] n n k p i j c p f " U e j g p g e v c f { " K p v s g U S A ; 9 4 . 2 % : [N P (S)] n N M R a n a l y s i s o f t h e s e v . " V g z v y q " u c o r n g u " k p f k e c v g f " v j g o " v q " d g " c " o k z v w t g (n N P ; 9 8 %) was obtained from Lancaster synthesis (Morecambe, Lancashire, UK). The benzoate ester of NP(F), (NPB), was prepared by standard benzylation of the parent phenol (NP; Fluka). The structural integrity of the oily product was confirmed by NMR spectroscopy and elemental analysis. The benzoate ester of nNP (nNPB) was prepared by benzylation of the parent phenol (nNP; source as above). V j g " r t q f w e v " y c u " c " u . q h v " y j N M R and mass spectroscopy confirmed its structure and elemental analysis indicated that it e q p v c k p g f " 3 1 8 " o q n g e u w n t g c " f q k h q " n y " c] v g g u t v 0 " c 3 / 9 8 / . 5 3 u 7 / q z * { 3 2 was synthesized from estrone by the method described by Francois and Levisalles (1968). C h v g t " t g e t { u v c n n k u c v k q p " h] t H t o c " p e ± { q e k n u q " j c g p z f c " p N g g " x k k v u " o r " 3 5 % } . / N M R and mass spectroscopy confirmed its structure and elemental analysis indicated that it contained 1/10 molecule of water. The corresponding benzoate derivative, 3 9 / f g u q z { g u v t c f k q n " d g p | q c v g "] g u v t B 3 / w a s . 5 . 7 / * prepared by a standard benzylation reaction. The product was recrystallized from o g v j { n c v g f " u r k t k v u " * 9 6 " Q R + " i k . x k N M R , " m a s s q n q w t n g

spectroscopy and elemental analysis confirmed the structure and purity of the product. The f k r j g p q n " f g t k x c v k x d g k " u q * h r " / p g f j q z { e j g q t h + 3 .. 3 / 3 / diphenol; 99%] was a gift from MD Shelby, originally supplied by Cedra Corp., Austin, Texas, USA.

Animals

K o o c v w t g " h g o c n g " C n r d n e x c p t w h e r e i n d i c a t e d o t h e r w i s e) w i t h b o d y { u " q n y g k i j v u " 5 : / 6 : i " " y g t g g u n i t a t v Z e n k a , A l d e r l e y P a r k . I n m a t u r e g " d t g g A l d e r l e y P a r k S w i s s A l b i n o m i c e w e r e o b t a i n e f " h t q o " v j g " u c o g " u q w t e g " * 4 y g k i j v u " 3 4 / 3 9 i + 0 " " C p k o c n u " y g t g " j q w u g f " k p " y k w a s c o n t r o l l e d a n d a 1 2 h / 1 2 h l i g h t d a r k c y c l e w a s m a i n t a i n e d . A n i m a l s w e r e w e a n e d o n H a r l a n T e k l a d T R M d i e t (H a r l a n U K , B i c e s t e t . " Q z h q t f u j k t g . " W M + " c v " 3 m a i n t a i n e d o n p e l l e t e d P C D d i e t (S p e c i a l D i e t S e r v i c e s L t d , W i t h a m , E s s e x , U K) f r o m 2 1 d a y s o n w a r d s . D i e t a n d w a t e r w e r e a v a i l a b l e a d l i b i t u m . A r a n g e o f a d d i t i o n a l d i e t s w e r e u s e d i n t h e r a t d i e t a r y s t u d y , a s d e s c r i b e d i n t h e l e g e n d t o F i g u r e 9 . R a t a n d m o u s e N o . 3 B r e e d i n g d i e t (R & M N o . 3) , r a t a n d m o u s e N o . 1 m a i n t e n a n c e d i e t (R & M N o . 1) a n d c a s e i n d i e t (S u p e r C a s e i n b a t c h R H M 9 2 / 1 1 7) w e r e a l s o o b t a i n e d f r o m S p e c i a l D i e t S e r v i c e s L t d . A l l a n i m a l s w e r e a c c l i m a t i z e d f o r 2 4 h b e f o r e b e i n g d o s e d .

Uterotrophic assays

The protocol for the uterotrophic assays was based on that described by Wakeling *et al.*, (1991). Animals were dosed by either oral gavage or subcutaneous (sc) injection. Test agents were dissolved or homogeneously suspended in arachis oil (except where indicated otherwise). The dosing volume for both routes of exposure was 5ml/kg body weight, except for the assays of coumestrol where a dosing volume of 10ml/kg was used. Animals received 3 daily doses of the test compound and were killed by an overdose of halothane 24h after the final dose. The dose levels shown in the Figures are the daily dose levels. Vaginal opening (or otherwise) was recorded at the time of death. Uteri were excised, trimmed free of fat, pierced and blotted to remove excess fluid. The body of the uterus was cut just above its junction with the cervix and at the junction of the uterine horns with the ovaries. The uterus was then weighed (wet weight). Uterine dry weight was also determined by drying the uteri at 70° h q t " 4 6 j " d g h q t g " e t a l . / 1 9 8 8 . D o s e l e v e l s w e r e u s u a l l y s e l e c t e d b a s e d o n p r e v i o u s f i n d i n g s f o r e s t r a d i o l , a n d i n t h e c a s e o f t h o s e c h e m i c a l s g i v i n g p o s i t i v e u t e r o t r o p h i c r e s p o n s e s , t h e m a x i m u m t o l e r a t e d d o s e (M T D) w a s p r o b a b l y n o t a c h i e v e d . T h e t o p d o s e l e v e l o f N P e v a l u a t e d (2 8 5 m g / k g) w a s t h e M T D (d a t a n o t s h o w n) . T h e a n a l o g s o f N P w e r e e v a l u a t e d t o t h i s s a m e t o p d o s e l e v e l . T h e t o p d o s e l e v e l o f c y p r o t e r o n e a c e t a t e g x c n w c v g f " * 4 0 7 " o i 1 m i + " y c u " e 7 / 3 2 2 " v k o g u " v j g " j

E₂ or E₂B, was used as a positive control agent in all of the studies, and each study was accompanied by a vehicle (negative) control group. Uterus wet weights are shown in the Figures, but uterus dry weights and uterus weight/body weight data were derived routinely and are discussed in the text. Generally, five animals per exposure group were employed, but this number was increased in some cases (see Figures).

Yeast assays

The yeast assay for estrogenic activity of chemicals, as described by Routledge and Sumpter (1996), has the DNA sequence of hER incorporated into the yeast genome. The molecular construct was designed to allow the activated" t g e g r v q t " v q " d k p f " v q " g sequences contained within a strong promoter region of an expression plasmid. When an active ligand binds to the estrogen receptor a *lacZ* reporter gene expresses the enzyme / galactosidase, which itself is quantitated using a chromogenic substrate. The methods used

were exactly as described earlier (Routledge and Sumpter, 1996). Briefly, the test chemicals were dissolved in ethanol and added to 96 well plastic plates at the concentrations shown in Figure 15. The ethanol was allowed to evaporate before the addition of medium containing the yeast. The plates were incubated at 32 °C for 4 days and quantitated in terms of the absorbance of wells at 540 nm. Vehicle and estradiol controls accompanied each experiment, and representative data are shown.

RESULTS AND DISCUSSION

The results and discussion are presented in relation to the three topics evaluated in this study. Chemical structures are shown in Scheme 1.

Critical protocol features of the uterotrophic assay

Estradiol benzoate (E_2D) increased uterine weights in animals 27 days at the start of the experiment and failed to respond further to the estrogenic challenge (Figure 1). Vaginal opening showed the same trend. The uterus weights 47 days at the end of the experiment were identical (Figure 1) despite a 20% difference in body weight at termination (data not shown). These observations indicate that uterotrophic assays based on absolute uterus weight is probably a better index of uterine growth than the uterus/body weight index (see later for extension of the latter point).

Comparisons of the oral and sc routes of administration of the three established estrogens estradiol (E_2), estradiol benzoate (E_2D), and estradiol valerate (E_2V) showed that E_2 is more active in the uterotrophic assay when injected subcutaneously, but a similar maximal effect can be achieved following oral administration of 400 g/kg of the endogenous estrogen E_2 was adopted as the standard positive control agent for the uterotrophic assay. The relationship between uterus wet weights, uterus wet weights relative to body weight, and uterus dry weights was essentially constant for all of the animals in the present study, and representative data for animals exposed to E_2 are shown in Table 1. Uterus wet weights expressed relative to body weights are subject to variation (Ashby *et al.*, 1997c), as illustrated by the slightly larger uterus/body weight ratio for animals exposed to the highest s.c. dose of E_2 , an effect dependant upon a small reduction in body weight for these animals (Table 1). We decided to regard absolute uterus wet weights as the primary assay parameter, with recourse to the determination of uterus dry weights in cases where it was considered necessary to confirm that a weak uterotrophic assay response was not due totally to increased water imbibation (c.f. such a suggestion; Evans *et al.*, 1996).

The enhanced sensitivity of the uterotrophic assay to E_2 when using sc injection, as opposed to oral administration, was also observed for both E_2D and E_2V . Subcutaneously administered EE was the most potent estrogen encountered in the present study, showing trophic activity at 0.5 g/kg. The choice of route of administration for chemical studies will be influenced by the purpose for which the data are to be used. Oral administration is generally employed in chemical toxicity studies to be employed for human risk assessments, and it seems reasonable to stay with this route for routine toxicity studies. In some cases, such as when the shape of the dose response relationship is the critical issue, it may be appropriate to employ parenteral administration, but it would be inappropriate to use such routes routinely, especially if the only justification for that is that these routes have been

used historically. The specificity of the uterotrophic assay for estrogens was indicated by the pgvkxg " t gur qpug " qdugt xgf " cv " vjg " fqug " ngxgn (Neumann, 1977; Figure 5).

The observation that premature vaginal opening appears to be a less sensitive marker of estrogenic activity than does the stimulation of uterine growth (Reel *et al.* 1996) has been eqphktogf " d { " vjg " r t In particular, premature vaginal opening does not always parallel increases in uterine weight. For example, at doses giving approximately equivalent uterotrophic effects, orally administered E₂ and EE produced premature vaginal qrgpkpi " yjkn " vjgkt " did not. To date, none of our control rats pkuvt c has had an open vagina at the time of thg " wvgt qvt q r j ke " cuuc { " * r q nonetheless, some chemicals produce a clear uterotrophic effect in the absence of vaginal opening (see coumestrol, later herein). This indicates that while premature vaginal opening can strengthen the classification of an agent as an estrogen, it cannot be taken as a definitive requirement.

The uterotrophic response produced by E₂ ycu " cdncv g f . " kp " c " fqug / t g n estrogen receptor antagonist ICI 182,780 (Wakeling and Valcaccia, 1983) when the latter was administered concomitantly with each dose of E₂ * Hkiwtg " 8 + 0 " Vjk u " kp f estrogens can be detected using the uterotrophic assay by the simple expedient of challenging the assay positive control agent with the test chemical. Animals exposed to ICI 182,780, with or without E₂, gave uterus weights that were decreased by ~50% when compared to concurrent control values. This led us to evaluate whether either the test vehicle, or the test diet, contained estrogenic contaminants. First, we evaluated whether the arachis oil vehicle was itself weakly estrogenic, some vegetable oils being known to contain phytoestrogens (Gunstone *et al.*, 1994). The data shown in Figure 7 confirm that there was no difference in the uterus weights of untreated, arachis oil treated, or distilled water treated rats. In each case, ICI 182,780 gave an approximately 50% reduction in uterus weights compared to concurrent controls. This indicated either that low levels of endogenous estrogens, or dietary phytoestrogens, were responsible for the elevated control uterus weights. The uterotrophic activity of the phytoestrogen coumestrol (Markarevich *et al.*, 1995) (Figure 8) confirmed that estrogenic dietary contaminants have the potential to influence control uterus weights in uterotrophic assays. The data shown in Figure 9 confirmed that dietary estrogens were the probable reason why our control uterus weights could be reduced by administration of ICI 182,780. Based on these data, the R&M no.3 / R&M no.1 diet combination was adopted for routine use with the assay. However, the experiments discussed in this paper were conducted before that change of diet occurred, thus the relatively high control uterus weights observed herein. Overall, the data shown in Figure 9 underline that diet should be considered as an important variable in any endocrine disruption assay, and that care should be taken to specify its source and constitution.

The rat was selected as the primary test species for the present studies. However, the mouse has been used equally in the past, and the general sensitivity and utility of the mouse uterotrophic assay was recently confirmed by Shelby *et al.* (1996). One of the xenobiotic estrogens evaluated by Shelby *et al.*, methoxychlor (MC), was found to be inactive in the two *in vitro* estrogenicity assays employed, probably because of the inability of these systems to remove the methyl groups of MC to yield the active diphenolic estrogenic species (Bulger *et al.* 1978). In fact, Shelby *et al.* (1996) demonstrated the latter diphenol to be estrogenic *in vitro*. In contrast, these authors reported MC to be uterotrophic to the mouse * uwd / ewvcpg. The data shown in Figure 10 reveal that MC shows similar uterotrophic activity in the rat and the mouse. " wukpi " gkvjgt " qtcn "

administration. Shelby *et al.* (1996) expressed their uterotrophic assay data corrected for body weight, and when the mouse data shown in Figure 10 are similarly corrected, MC is found to have given a positive response of similar magnitude (i.e., a 100% increase in uterus weights over concurrent control levels, at 500 mg/kg MC). These limited data for MC suggest that the choice of test species and route of chemical administration may only lead to quantitative, rather than qualitative, differences in uterotrophic assay outcomes. However, qualitatively different test results will inevitably emerge for chemicals that are differentially absorbed and/or differentially metabolized in different test species. The choice of test species and route of administration for the uterotrophic assay should be kept under review as new data accumulate.

Structural basis of the estrogenicity of nonylphenol

Commercial nonylphenol (NP) is a mixture of several isomers. The commercial material is known to possess estrogenic properties (Soto *et al.*, 1995; Routledge and Sumpter, 1997), and these have been suggested to arise due to the structural similarity to E₂ of one or more of the isomers of NP. Warhurst (1994) has suggested one particular isomer of NP as being the most likely to bind to and activate the estrogen receptor (Scheme 1). This isomer happens to be similar in structure to octylphenol, a related compound that contains only one isomer (as shown in Scheme 1) and which is active as an estrogen *in vitro* (Routledge and Sumpter, 1997) and *in vivo* (Bicknell *et al.*, 1995). Recently, Sanborn (1996) erroneously associated the estrogenic properties of commercial NP with the linear aliphatic sidechain isomer nNP, a compound which is not estrogenic (see below). Sanborn (1996) also suggested that hydroxylation of the aliphatic sidechain of nonylphenols (shown for nNP in Scheme 1) was required to complete their structural alignment with E₂. However, that suggestion is not supported by the observation that NP, *per se*, competes with E₂ in bindings to isolated cytosolic estrogen receptors, i.e., without the need for prior metabolic transformation (Soto *et al.*, 1995; White *et al.*, 1994).

As recently reported by Lee and Lee (1996), NP is a weak agonist in the immature rat uterotrophic assay (Figure 11). This response was abolished by the estrogen receptor antagonist ICI 164,384, thereby confirming that the effect was mediated *via* the estrogen receptor (Figure 11). Similar levels of uterotrophic activity were observed for the two different isomers of NP (Figure 12). The inactivity of the straight chain analog of NP (nNP; Figure 12) was supported by its inactivity in the yeast assay (see below). The inactivity of nNP favors the interpretation that the interaction of the chemical with the estrogen receptor (Routledge and Sumpter, 1997; Warhurst, 1994). It is interesting to note the weak assay responses for NP shown in Figure 12. The reproducibility of these weak effects, and the observation of cases of vaginal opening among the test animals, establish uterotrophic activity in these cases; nonetheless, these data indicate the urgent need to agree criteria for discerning activity in the uterotrophic assay (Ashby *et al.*, 1997c).

It was noted earlier that E₂B is a more potent estrogen in the uterotrophic assay than is E₂ itself (Figures 2 and 3). This enhancement of activity is probably caused by a combination of factors, such as reduced conjugative excretion of the free phenol (E₂) by protection of its acidic phenol substituent, and the enhanced lipophilicity of the benzoate (E₂B) in comparison to the parent phenol (E₂). It therefore became of interest to evaluate if the benzoate derivative of NP (NPB) would be more potent than NP itself, and in particular, if the benzoate derivative (nNPB) of the inactive straight chain analog (nNP) would be active in the uterotrophic assay. The data shown in Figure 13 demonstrate that these two benzoates

were of similar activity to the parent phenols; equally as positive, and negative, respectively. These rather surprising findings indicate that the activity of NP cannot be generalized, and that nNP seems to be devoid of intrinsic estrogenic activity.

The suggestion that the estrogenic activity of NP is due to the ability of some of its constituent isomers to mimic estradiol in binding to the estrogen receptor (Routledge and Sumpter, 1997; Warhurst, 1994) indicates that the 17 β -estradiol-like isomers of NP should represent the progenitor 'estrogenic' activity. It is reported to be uterotrophic to hypophysectomised rats (Huggins and Jensen, 1955) and to be able to compete with estradiol in binding to isolated estrogen receptors (Fanchenko *et al.*, 1993; 9; +0). NP was uterotrophic to immature rats (Figure 14), showing a slightly higher minimum detection level than E₂ and a more flattened dose response relationship (Figures 3 and 14). As reported in the uterotrophic assay to vjcvqh (Figure 14). It therefore appears that the suggestion that one or more of the branched chain isomers of NP is able to mimic estradiol in binding to the estrogen receptor is a valid and useful observation. However, the suggestion that the aliphatic sidechain of NP requires to be hydroxylated before NP acquires estrogenic activity (Sanborn, 1996) seems to be unlikely.

Relative sensitivity of the yeast estrogenicity assay and the uterotrophic assay

In vitro assays will be of particular value in cases where a lead chemical showing estrogenic activity is shown to be positive in the *in vitro* assay, and where the activity of structural analogs of the lead chemical are the subject of study. The yeast hER assay described by Routledge and Sumpter (1996), has shown sensitivity to a wide range of chemical classes of estrogens, and the assay has intrinsic metabolic competence, as evidenced by its ability to detect as positive the estrogen MC (Figure 15; Ashby *et al.*, 1997b), despite the inability of some other *in vitro* assays to detect it (Shelby *et al.*, 1996). This attribute of the yeast assay is probably due to the enhanced metabolic competence of growing yeast cells and their presumed ability to demethylate MC to the active estrogenic diphenol derivative (OE / fkrjgpetal., 1996; Burger *et al.*, 1978). As expected, the yeast assay was more sensitive to the ultimate estrogenic species (fgtkxgf h tq o) OE * OE / fkrjg to MC itself (Figure 15).

Kp vjg { gcuvz" y u u c { c.d"q fw g ü qv zg { p // G k₂ (Figure 15). This reduced activity is consistent with vjg t g n c v k x g " c₂ and kEx in the gw v g t q v t q r j k e " c u u c { " * H k i w t g u " 3 6 " ₂ *in vitro* and *in vivo* has implications for the derivation of structure activity relationships among xenobiotic estrogens (Katzenellenbogen, 1995; Waller *et al.* " 3 ; ; 8 + " y j g t g " v j g " r t g u g hydroxy and the 17 / j { f t q z { " i t q w r u " q considered to be necessary for t g " i g p estrogenic activity.

The yeast assay is less sensitive to E₂B than it is to E₂ (Figure 15). This is in contrast to the marginally enhanced activity of E₂B, compared to E₂, in the uterotrophic assay (Figures 2 and 3). This indicates that the yeast assay is r t q d c d n { " c d n g " v q " j { f t q n { u group present in E₂B, but that this is not completed under the conditions of the test. As observed with the uterotrophic assay, the yeast assay found the two different samples of NP to be of similar estrogenic activity, and nNP to be inactive (Figure 15). In surprising contrast to in the uterotrophic assay, the benzoate derivative of NP (NPB) was inactive in the yeast

assay (Figure 13). This suggests that the yeast cells possess only selective esterase activity, being able to hydrolyse partially the benzoate ester of E₂B, while being unable to hydrolyse the benzoate ester of NPB.

In summary, the yeast estrogenicity assay has demonstrated high general sensitivity, and a level of metabolic competence, in the present studies. A potential weakness of this (and probably of all other *in vitro* assays) is highlighted by the failure of the yeast assay to detect as positive the *in vivo* estrogen NPB. That failure is perhaps reflected quantitatively by the reduced sensitivity of the assay to E₂B in comparison to E₂. Although the problem posed by the inadequate metabolic competence of the current *in vitro* estrogenicity assays must be solved before they can achieve their true potential (Ashby *et al.*, 1997a; Ashby, 1997), the partial metabolic competence of the present yeast assay confirms it as a valuable screening test.

CONCLUSIONS

The results described here illustrate the intrinsic complexity of evaluating chemicals for estrogenic activities. Suitable detection systems exist, but the manner in which they are deployed can influence the toxicological value of the data derived. It is recommended that when designing experiments in this area of toxicology it is important consider the purposes for which the test data will be used. This will influence practical decisions, such as choice of *in vitro* or *in vivo* assays, and the choice of dose levels and route of chemical administration in rodent studies. At the practical level, there is an urgent need to agree criteria for assessing the activity of chemicals in the available estrogenicity assays, and to enhance the metabolic competence of the currently available *in vitro* assays.

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REFERENCES

Ashby J. (1997). Endocrine modulation of human reproduction by environmental chemicals. *Environ. Toxicol. Pharmacol.* in press.

Ashby J., Houthoff E., Kennedy S.J., Stevens J., Bars R., Jekat F.W., Campbell P., Van Miller J., Carpanini F.M., and Randall G.L.P. (1997a). The challenge posed by endocrine disrupting chemicals. *Environ. Hlth. Perspect.* **105**, 22 / 220 "

Ashby J., Lefevre P.A., Odum J., Harris C.A., Routledge E.J., and Sumpter J.P. (1997b). Synergism between synthetic oestrogens? *Nature* **385**, 494.

Ashby J., Odum J., Tinwell H., and Lefevre P.A. (1997c). Assessing the risks of adverse g p f q e t k p g / o g f k c v g f " g Reg. Tox. Pharmacol. in press. g " v q " h t q

Bicknell R.J., Herbison A.E., and Sumpter J.P. (1995). Oestrogenic activity of an environmentally persistent alkylphenol in the reproductive tract, but not the brain of rodents. *J. Steroid Biochem. Molec. Biol.* **54**, 9 / ; 0

D t c p j c o " U 0 . " N g c o q p u " O 0 N 0 . " c p f " U j g g j c p " F 0 O 0 " * ornithine decarboxylase activity and uterine growth in the rat. *J. Steroid Biochem.* **29**, 375 / 37 ; 0

Bulger W.H., Muccitelli R.M., and Kupfer D. (1978). Studies on the *in vivo* and *in vitro* estrogenic activities of methoxychlor and its metabolites. *Biochem. Pharmacol.* **28**, 4639 / 4645 0

- functional development. Princeton Scientific Publishing.
- Evans G.L., Bryant H.U., Magee D.E., and Turner R.T. (1996). Raloxifene inhibits bone turnover and prevents further cancellous bone loss in adult ovariectomized rats with established osteopenia. *Endocrinology* **137**, 635; / 63660
- Fanchenko N.D., Sturchak S.V., Shchedrina R.N., Pivnitskii K.K., Novikov E.A., and Ishkov V.L. (1979). The specificity of the human uterine receptor. *Acta Endocrinol. (Copenhagen)* **90**, 389 / 3970
- Hutchinson R.O. (1996). Estrogenic activity of 2,4,6-trichlorophenol and 2,4,6-trichlorophenol with oxygenated functions at positions 6 or 16: The impeded estrogens. *J. Expt. Med.* **102**, 557 / 5680
- Katzenellenbogen J.A. (1995). Commentary: The structural pervasiveness of estrogenic activity, *Environ. Hlth. Perspect.* **103 (Suppl.7)**, ; ; / 3230
- Lee P.C., and Lee W. (1996). *In vivo* estrogenic action of nonyl phenol in immature female rats, *Bull. Environ. Contam. Toxicol.* **57**, 563 / 56 : 0
- Markarevich B.M., Webb B., Densmore C.L., and Gregory R.R. (1995). The effects of coumestrol on estrogen receptor function and uterine growth in ovariectomized rats. *Environ. Hlth. Perspect.* **103**, 796 / 7 : 30
- McLachlan J.A. (1993). Functional toxicology: a new approach to detect functionally active xenobiotics. *Environ. Hlth. Perspect.* **101**, 5 : 8 / 5 : 90
- McLachlan J.A., and Korach K.S. (1995). Estrogens in the environment III: global health implications. *Environ. Hlth. Perspect.* **103 (Suppl.7)**, 5 / 39 : 0
- Neumann F. (1977). Pharmacology and potential use of cyproterone acetate. *Horm. Metab. Res.* **9**, 3 / : 0
- O'Connor J.C., Cook J.C., Craven S.C., Van Pelt C.S., and Obourn J.D. (1996). An *in vivo* battery for identifying endocrine modulators that are estrogenic or dopamine regulators. *Fund. Appl. Toxicol.* **3**, 3 : 4 / 3 ; 70
- Reel J.R., Lamb J.C., and Neal B.H. (1996). Survey and assessment of mammalian estrogen biological assays for hazard characterization. *Fund. Appl. Toxicol.* **34**, 4 : : / 5270
- Routledge E.J., and Sumpter J.P. (1996). Estrogenic activity of surfactants and their degradation products assessed using a recombinant yeast screen. *Environ. Toxicol. and Chem.* **15**, 463 / 46 : 0
- Routledge E.J., and Sumpter J.P. (1997). Structural features of alkylphenolic chemicals associated with estrogenic activity. *J. Biol. Chem.* **272**, 54 : 2 / 54 : : 0
- Sanborn J.R. (1996). An hypothesis for estrogenicity of alkylphenols and the possible role of metabolism. *Pest. Outlook Oct.*, 58 / 590
- Shelby M.D., Newbold R.R., Tully D.B., Chae K., and Davis V.L. (1996). Assessing environmental chemicals for estrogenicity using a combination of *in vitro* and *in vivo* assays. *Environ. Hlth. Perspect.* **104**, 34 ; 8 / 35220
- Soto A.M., Sonnenschein C., Chung K.L., Fernandez M.F., Olea N., and Serrano F.O. (1996). Estrogenic activity of environmental pollutants. *Environ. Hlth. Perspect.* **103 (Suppl 7)**, 335 / 3440
- Wakeling A.E., Dukes M., and Bowler J. (1991). A Potent Specific Pure Antiestrogen with Clinical Potential. *Cancer Res.* **51**, 5 : 89 / 5 : 950
- Wakeling A.E., and Valcaccia B. (1983). Antiestrogenic and antitumour activities of a series of 4-hydroxyphenyl derivatives. *J. Endocrinol.* **99**, 666 / 6860

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Chem. Res. Toxicol. **9**, 3 4 6 2 / 3 4 6 : 0

Warhurst A.M. (1994). An environmental assessment of alkylphenol ethoxylates and
c n m { n r j g p q n u 0 " r r 3 / 3 7 . " H t k g p f u " q h " v j g " G c t v

White R., Jobling S., Hoare S.A., Sumpter J.P., and Parker M.G. (1994). Environmentally
persistent alkylphenol compounds are estrogenic. *Endocrinology* **135**, 3 9 7 / 3 : 4 0

Table 1. The effect of 17 / g u v t c f k q n " q p " f k h h g t g p v " r c t c o g v g t administered by either oral gavage or subcutaneous (sc) injection. Group sizes are as described in Figure 2.

Estradiol g/kg	Mean body weight (g; SD)	Uterus wet weight (mg)		Uterus wet weight (mg)/ body weight (g)		Uterus dry weight (mg)			
		mean SD	% of control values	mean SD	% of control values	mean SD	% of control values		
Oral gavage	0	63.5 4.6	33.7 8.1	100	0.531 0.13	100	6.69 1.06	100	
	10	61.7 1.8	37.5 7.5	112	0.609 0.12	115	6.30 0.85	94	
	20	62.8 4.0	39.0 7.4	116	0.624 0.13	117	6.99 1.16	104	
	40	60.9 3.6	38.3 4.3	114	0.630 0.07	119	7.66 0.84	114	
	100	60.8 3.4	68.1 17.5	202	1.120 0.29	211	12.15 3.20	182	
	200	60.7 3.2	70.9 14.9	210	1.165 0.22	219	13.05 3.70	195	
	400	63.2 5.4	101.7 16.7	302	1.614 0.26	304	19.37 2.36	289	
	sc injection	0	63.8 4.8	38.2 7.3	100	0.597 0.10	100	7.09 1.13	100
		0.5	66.0 3.2	40.5 3.2	106	0.616 0.06	103	7.83 0.65	111
1		63.4 3.7	42.3 7.0	111	0.666 0.08	112	8.10 1.29	114	
2		65.3 4.8	43.9 5.8	115	0.670 0.04	112	8.34 1.01	118	
5		63.1 5.2	86.4 9.8	226	1.368 0.11	229	15.00 2.21	212	
10		62.7 3.1	95.4 17.0	250	1.519 0.25	254	16.92 2.31	239	
20		61.8 3.6	86.4 13.4	226	1.482 0.12	248	15.73 2.32	222	
40		63.1 2.9	84.9 6.0	222	1.348 0.09	226	15.61 1.28	220	
200		60.6 3.6	89.0 11.8	235	1.486 0.20	249	16.45 1.65	232	
400		61.9 4.5	101.2 14.5	265	1.628 0.16	273	17.98 2.31	254	

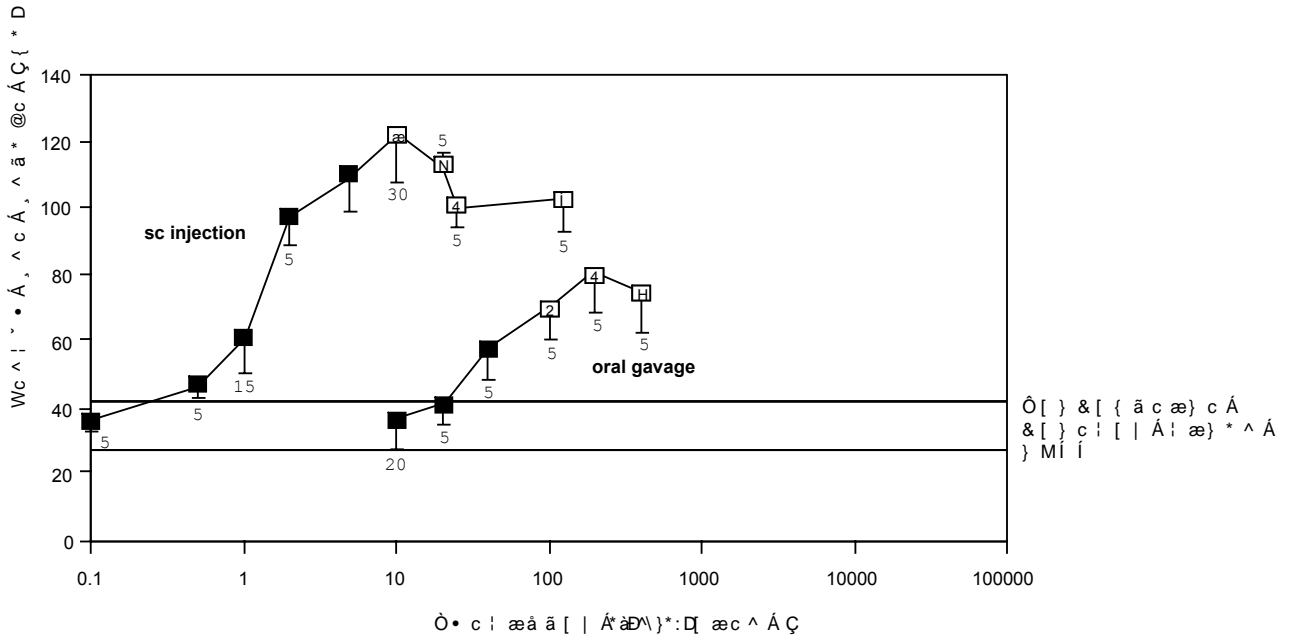


FIGURE 3 È Á V @ ^ Á ^ ~ È ^ & cc ! Á ð á - Á [F I Á á ^ } : [æ c ^ Á [] Á ^ c ^ ! ~ • Á, ^ c Á, ^ á * @c Á Ç { * D } á Á ç æ * á } æ | Á [] ^ } á Ó [] c ! [] • Á ! ^ & ^ á ç ^ á Á æ ! æ & @ á • Á [á | Á [] | ^ È Á Ó [] • á Á • ^ { à [] • Á á ^ } [c ^ Á] [Á æ] á { æ | Á, á c @ Á [] ^ } Á ç æ * á } á { æ | Á, á c @ Á [] ^ } Á ç æ * á } á Á Ç c @ ^ Á } { á | Á Ç Á - Á [] ^ } Á ç æ * á } æ • á • Á • @ [] • á c @ Á [] ^ } Á ç æ * á } á } á * Á [c Á á ^ c ^ ! { á } ^ á È Á Ó æ c æ Á ! ^ } ! ^ • ^ } c Á * ! [~] Á { ^ æ } • Á † Ú Ç & [{ á á } ^ á Á ~ ! [{ Á J Á á } á ^ } ^ } á ^ } c Á • c ^ á á ^ • È Á ^ æ & @ Á [~ Á, @ á & @ Á & [] c æ á } ^ á Á æ Á ç ^ @ á & | ^ Á á ^ [] , Á c @ ^ Á á æ ! Á ! ^ } ! ^ • ^ } c • Á c @ ^ Á & [{ á á } ^ á Á * ! [~] Á • á : ^ È Á Á

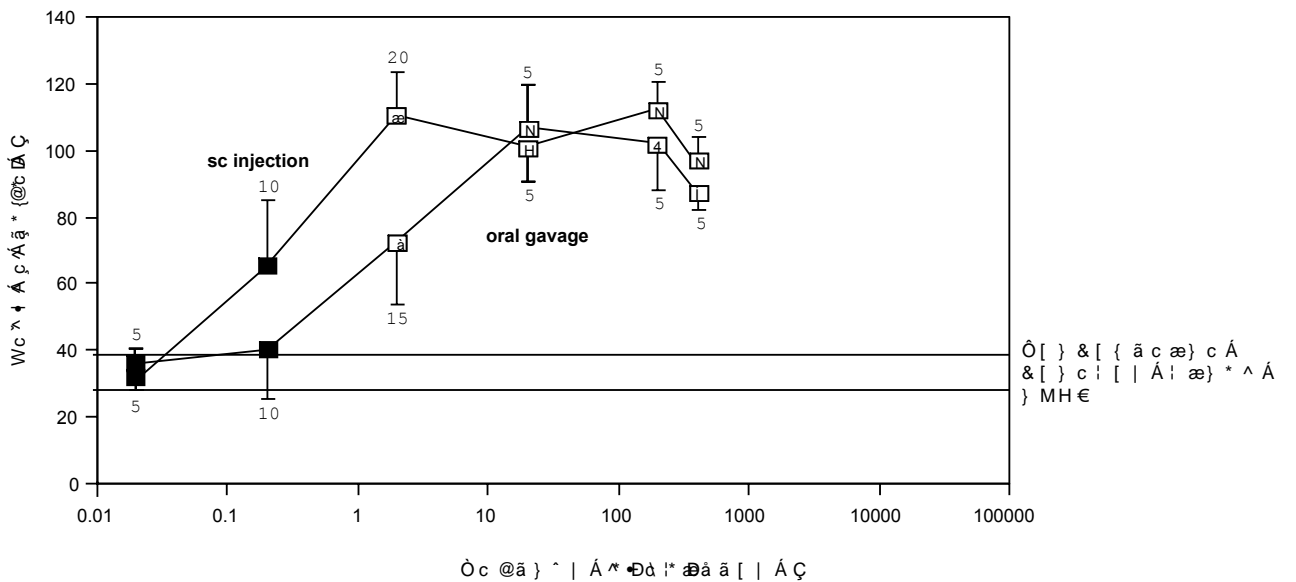


FIGURE 4 È Á V @ ^ Á ^ ~ ~ ^ & c • Á [~ Á ^ c @ á } á * ! @ c Á Ç { * D } á Á ç æ * á } á Á ç æ * á } á Á Ç c @ ^ Á } { á | Á Ç Á - Á [] ^ } Á ç æ * á } æ • á • Á • @ [] • á c @ Á [] ^ } Á ç æ * á } á } á * Á [c Á á ^ c ^ ! { á } ^ á È Á Ó æ c æ Á ! ^ } ! ^ • ^ } c Á * ! [~] Á { ^ æ } • Á † Ú Ç È Á X æ | ^ • Á , ^ ! ^ Á & [{ á á } ^ á Á ~ ! [{ Á J Á á } á ^ } ^ } á ^ } c Á • c ^ á á ^ • È Á ^ æ & @ Á [~ Á, @ á & @ Á & [] c æ á } ^ á Á æ Á ç ^ @ á & | ^ Á á ^ [] , Á c @ ^ Á á æ ! Á ! ^ } ! ^ • ^ } c • Á c @ ^ Á & [{ á á } ^ á Á * ! [~] Á • á : ^ È Á Á

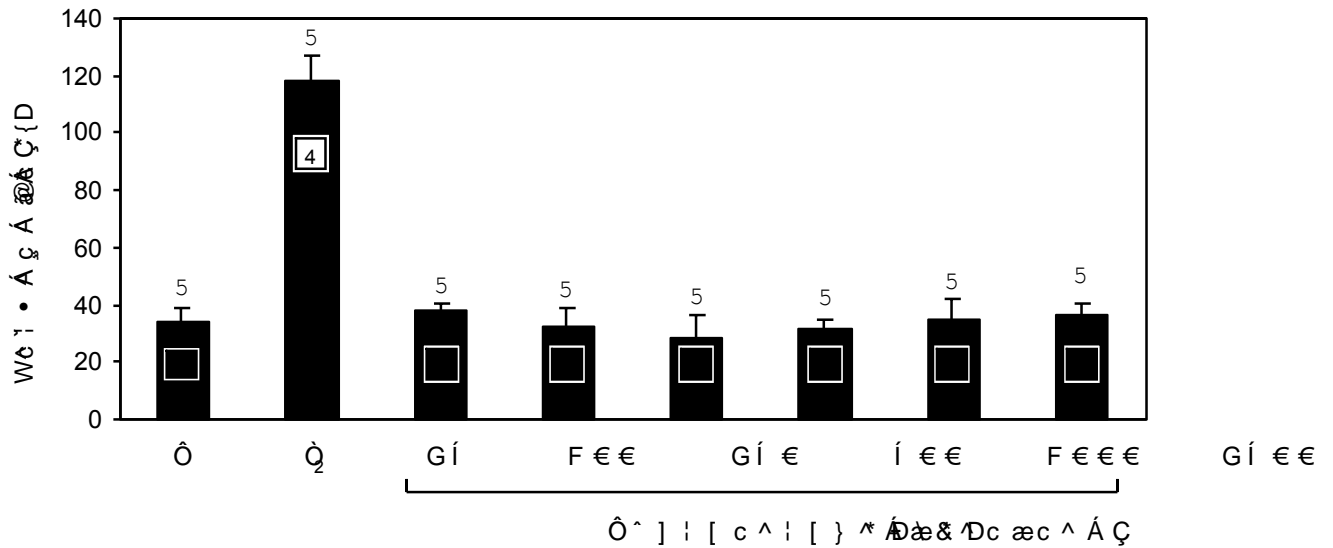


FIGURE 5 È Á Á V @ ^ Á ^ ~ ~ ^ & c · Á [~ Á Á [| æ | Á æ à { ã } ã · c | æ c ã [} Á [~ Á c @ ^ Á æ } c ã ^ c ^ | ^ · Á ^ c Á ^ ã * @ c Á æ } á Á ç , ð Á Ç } * ð Á [É] Á [] ; æ } * Á È Á ç Q · æ ^ Á ð Á ã [æ] · Á Ç Ö ^ á Á [] [· ã c ã ç ^ Á & [] c i [| É Á Ö [] c i [] · Á | ^ & ^ ã ç ^ á Á æ i [] & @ ã · Á [ã | Á [] | ^ É Á [] ^ Á æ } ã { æ | Á , ã c @ Á [] ^ } Á ç æ * ã } æ Á Ç } ^ { à ^ | Á [~ Á [] ^ } Á ç æ * ã } æ · Á · @ * i [^] Á { ^ æ } · Á † Û Ö È Á V @ ^ Á } ^ { à ^ | Á æ à [ç ^ Á c @ ^ Á à æ i Á | ^] | ^ · ^ } c · Á c

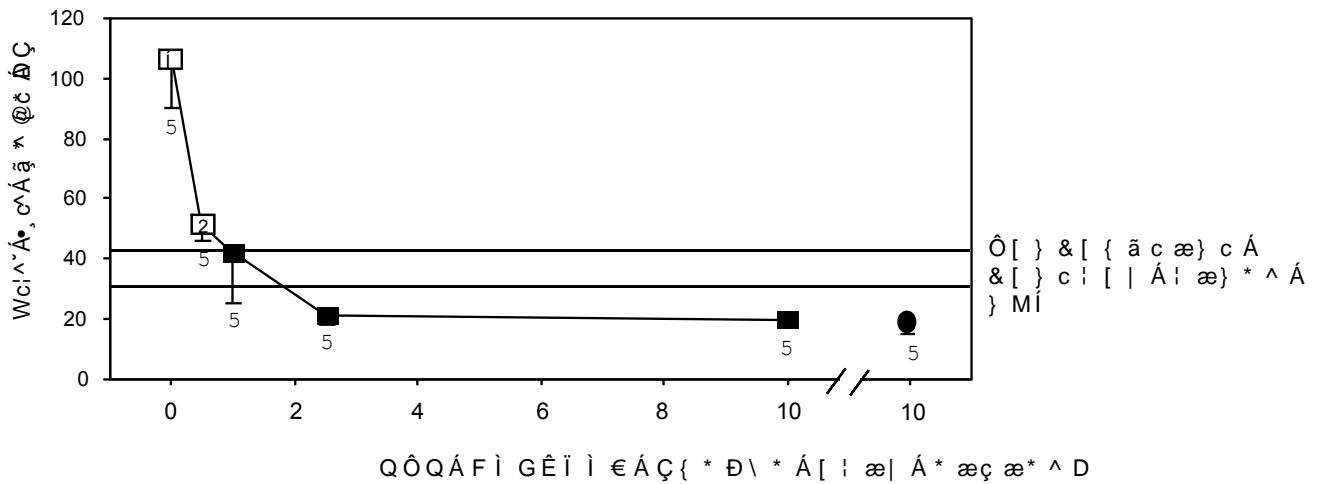
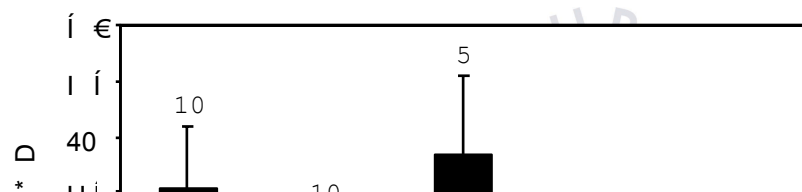


FIGURE 6 È Á V @ ^ Á ^ ~ ~ ^ & c · Á [~ Á c @ ^ Á ^ · c i [* ^ } Á | ^ & ^] c [| Á æ } c æ * [] ã · c Á Ç ^ c ^ | [c i [] @ ã & Á æ & c ã ç É Á c * Á [| ~ á | Á * c æ ç ã á ^ D Á Á Ç] Á ç { æ c ^ | ^ Á i æ c · É Á Ö [] æ } ã { æ | · Á | ^ & ^ ã ç ^ á Á æ i æ & @ ã · Á [ã | Á [] | ^ É Á V @ ^ Á ^ ~ ~ ^ & c · Á [~ Á Q Ô Q Á F i æ i ^ Á · @ [,] Á [] Á c @ ^ Á · æ * @ c Á [~ Á c @ ^ Á ~ ã * ^ i ^ Á Ç Á Á Á Á D È Á Ö [] · ^ á Á · ^ [] ^ } Á ç æ * ã } æ È Á [] ^ } Á · ^ { à [| · Á á ^ } [c ^ Á æ c Á | ^ æ · c Á [] ^ Á æ } ã { æ | Á , ã c @ Á [] ^ } Á ç æ * ã } æ · Á ã · Á · @ [,] Á , ã c @ ã } Á c @ ^ Á à æ c æ [ã } c Á à [c { ^ æ } · Á † Û Ö È Á V @ ^ Á } ^ { à ^ | Á à ^ | [, Á c @ ^ Á à æ i Á | ^] | ^ · ^ } c · Á c @ ^ Á * i [^



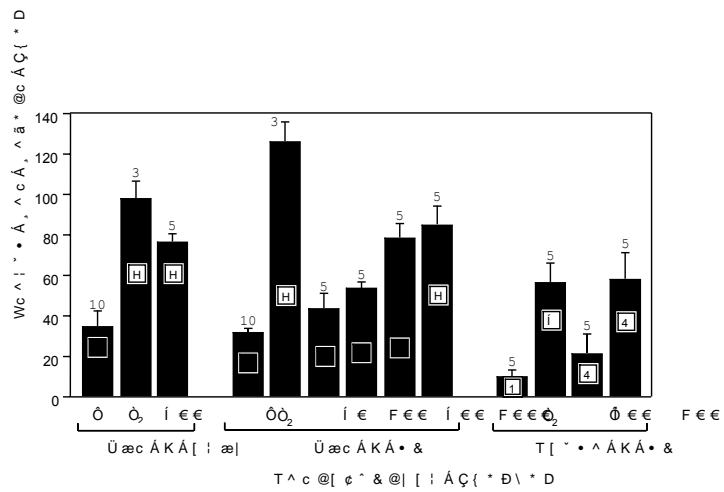
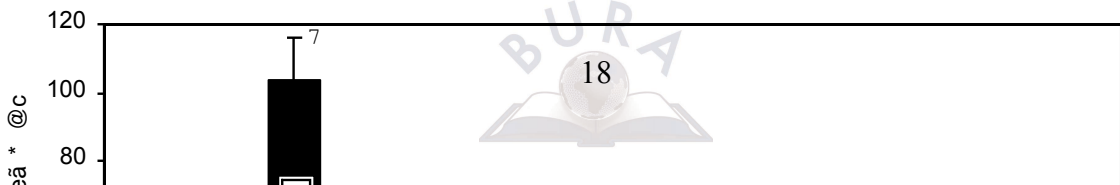


FIGURE 10: ... (The caption text is highly illegible and mirrored in the original image)



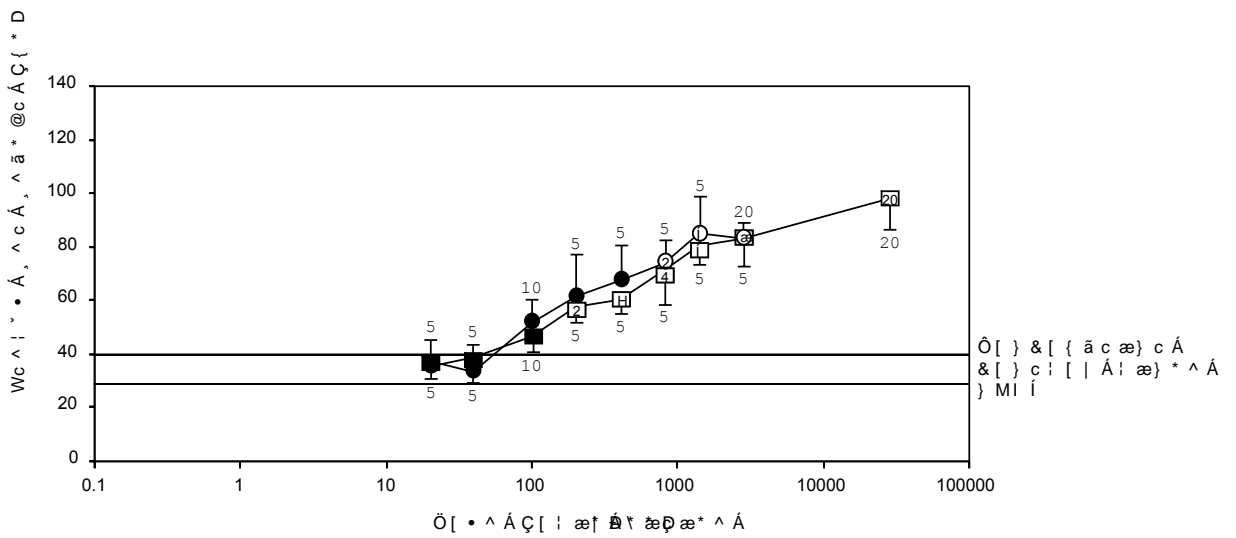
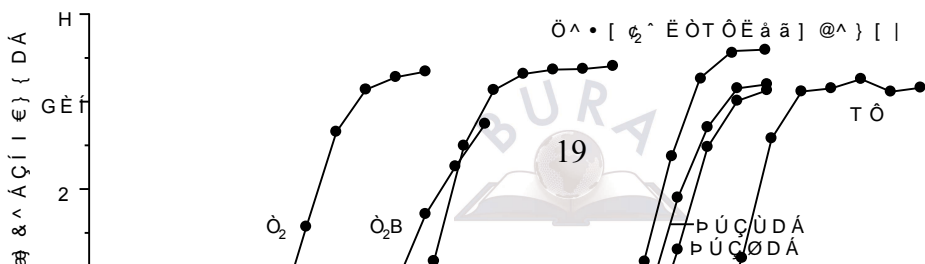


FIGURE 14 $\text{Wc}^{\wedge} \text{!}^{\cdot} \text{Á}, \text{^} \text{c} \text{Á}, \text{^} \text{ä}^* \text{@} \text{c} \text{Á} \text{Ç} \{^* \text{D}$
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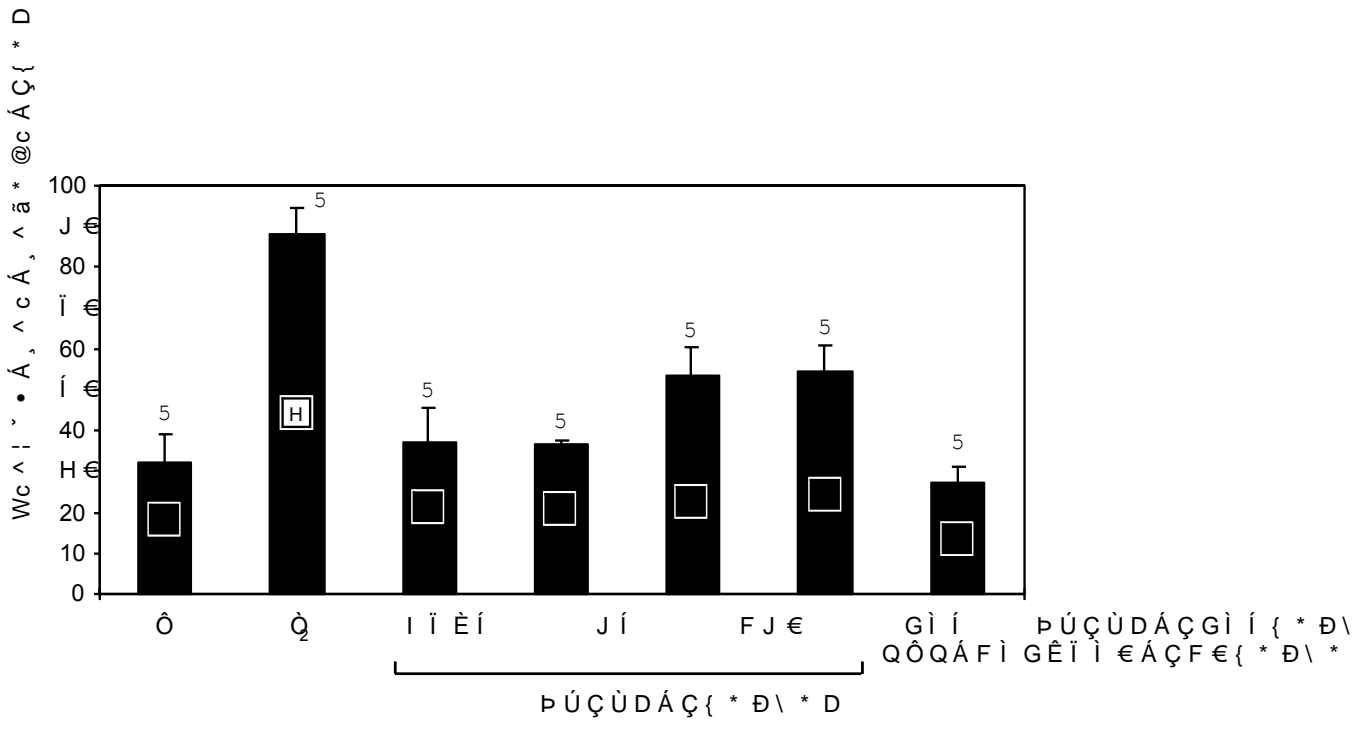


FIGURE 11 ÈÁÁV@^Á^ ~^&c •Á[~ÁÁ[| æ | Áæá { ā } ā • c | æc ā [] Á [~Áà : æ] &@^á Á
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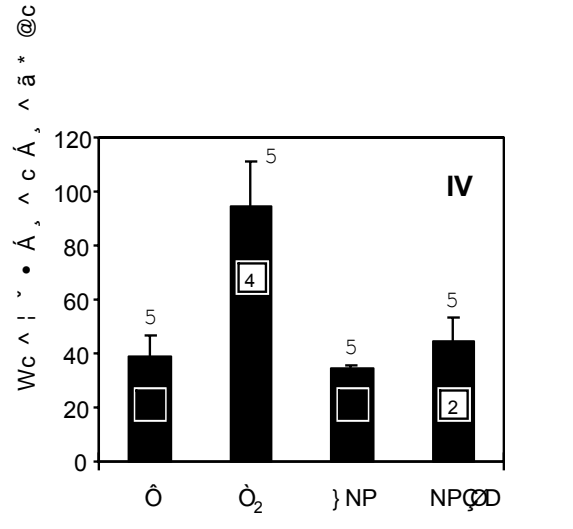
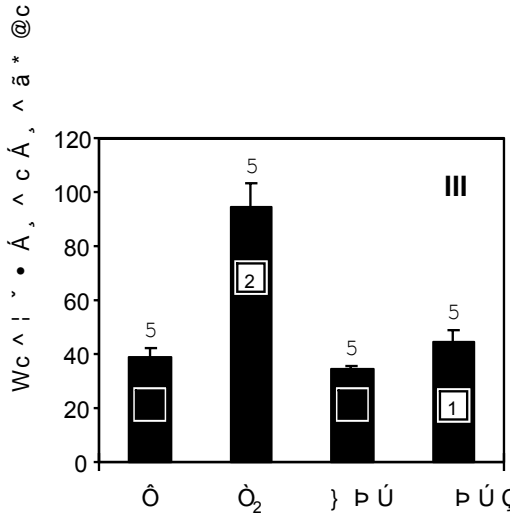
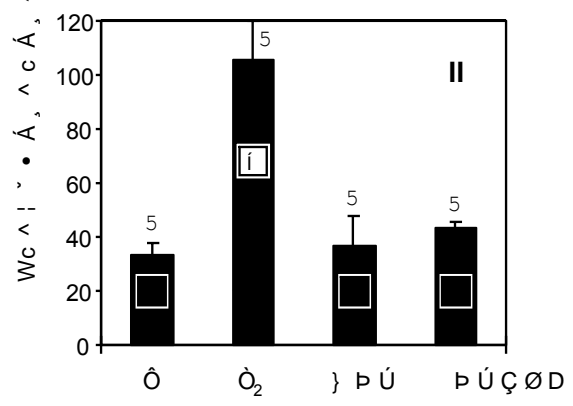
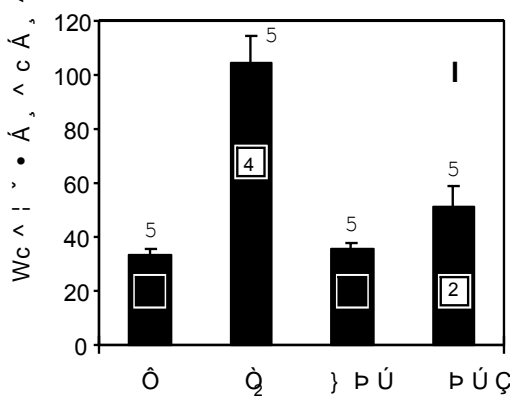


FIGURE 12 ÈÁÁV@^Á^ ~^&c •Á[~Ác @^ÁÁØ | ^ | æÁ • æ [] | ^ Á [~ Á à : æ] &@^á Á } [] ^ |
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