

# 1 Evaluation of the preparation and fertilizer release performance of planting concrete made 2 with recycled-concrete aggregates from demolition

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## 7 Abstract

8 This paper developed a new generation of planting concrete (PC) made from recycled aggregates of demolished  
9 concrete (RADC) to improve the fertilizer retention properties and reduce the cost. By optimizing the fabrication  
10 procedures and mix proportions of the planting concrete with RADC (PC-RADC), the interconnected porosity,  
11 water permeability coefficient and 28-day compressive strength can be enhanced to 40.9 %, 2.88 cm/s and 6.5 MPa,  
12 respectively, which laid a fundamental basis for the improved release of fertilizer in the developed planting  
13 concrete. The experimental results revealed that with the addition of 4.4 kg/m<sup>3</sup> urea fertilizer, not only the 28-day  
14 release rate of nitrogen, a key parameter representing the fertilizer release performance, of the PC-RADC was  
15 improved by 72.1 %, but also a more stable pore fluid alkalinity at approximately 8.20 pH was developed, which  
16 provided a suitable environment for plant growth. Moreover, the frost resistance of the urea fertilized PC-RADC  
17 was enhanced by 12.4 % with the reduction of 150-day drying shrinkage rate by 20.5 %. In conclusion, utilization  
18 of RADC has a great potential for the sustainable PC development with improved fertilizer release performance.

19 **Key words:** Planting concrete (PC); Recycled aggregates from demolished concrete (RADC); Fertilizer release  
20 performance.

21

## 22 1. Introduction

23 Utilization of porous concrete as a substitute for conventional concrete has been rapidly developed towards a  
24 sustainable, multifunction and intelligent material to realize the environment protection and sustainable  
25 development of modern civil infrastructure. Due to the enormous advantages of porous concrete, such as excellent  
26 water permeability, good thermal insulation properties and lightweight nature (Lu et al., 2015, 2018), it not only  
27 provides a realistic solution to erosion and flooding caused by impervious land development, but also satisfies the  
28 bio-adaptability and plant-growing characteristics of substrate as a planting concrete (PC) (Kim and Lee, 2010;  
29 Park and Tia, 2004; Jang et al., 2015; Ćosić et al., 2015), as shown in Fig. 1(a) (Huang et al., 2016; Li et al., 2017a;  
30 Lu et al., 2016). It also has been widely applied in permeable trenches, gullies, and gutters (Bhutta et al., 2012), and

31 in noise barriers (Ćosić et al., 2015; Kim and Lee, 2010) etc. In fact, PC is a ecological type of porous concrete,  
32 compared with ordinary concrete, it is more efficient to protect water and soil loss on the slopes of river banks  
33 and roadbeds (Fig. 1(b)) (Tarnai et al., 2004, 1998) and therefore prevent the soil erosion and desertification (Li et  
34 al., 2017a; Yan et al., 2013). Roots of plants could pass through the PC via pores structure to absorb nutrition  
35 underground (Fig. 1(c)). However, in order to ensure efficient plant growth in PC, the following issues should be  
36 addressed: (1) providing enough spaces for root growth; (2) ensuring the effective retention of fertilizer and (3)  
37 reducing the amount of alkali leaching (Li et al., 2017a; Tarnai et al., 2004).

38 Regarding the issue of providing enough spaces for root growth, Tarnai et al. suggested that the interconnected  
39 void ratio of PC should be at least 25 % for immediate growth, 21 % for late growth and 18 % for viable growth  
40 (Tarnai et al. 2004). Besides, it also revealed that the addition of coarse aggregates with the range of 15-20, 15-25,  
41 20-25 and 20-30 mm benefited to developing PC with suitable interconnected void ratio. Among various of coarse  
42 aggregates, crushed natural resources, including limestone and granite, have been regarded as an excellent  
43 candidate to fabricate PC due to its low water absorption rate and crush index et al., but the natural resources crisis  
44 has greatly limited its widely application (Taylor et al., 2007; Shi et al., 2016). Therefore, it is of importance to find  
45 a sustainable replacement material in large quantities with technical performance comparable to that of existing  
46 natural aggregates. Currently, many waste materials have been used to replace natural aggregates, for instance steel  
47 slag, copper slag and construction and demolition waste. Among them, construction and demolition waste has been  
48 assessed as a reliable source of untapped waste materials with a high valorisation potential and the most significant  
49 waste streams in the world, accounting for over 650 million tonnes per year, with waste demolition concrete (WDC)  
50 representing 85 % (Silva et al., 2017; João et al., 2017). Therefore, utilization of WDC as recycled-concrete  
51 aggregates for producing PC would offset the energy intensive quarrying process of natural aggregates and protect  
52 the environment from waste landfill. More importantly, it is very convenient and cost-effective for post-processing  
53 (breaking, removing and crushing) of the WDC for recycling aggregates use (Blengini and Garbarino, 2010;  
54 Miguel et al., 2015). All the benefits above have driven the WDC as an energy effective materials widely utilized in  
55 various applications, such as bulk filling, revetment, roadbeds, base material for drainage structures, noise barriers  
56 and embankments (Ajdukiewicz and Kliszczewicz, 2002; Silva et al., 2017). Although extensive research has been  
57 conducted to investigate the influence of waste demolition concrete as recycled aggregates on the properties of  
58 porous concrete (Ćosić et al., 2015; Lamond et al., 2002; Bhutta et al., 2013;), few of them studied the efficiency of  
59 recycled-concrete aggregates from demolition for PC.

60 Regarding the issue of fertilizer retention of PC, solid or liquid forms of fertilizer should be regularly supplied  
61 since nutrients content (e.g. nitrogen and phosphorus) necessary for plant growth is scarce in Portland cement  
62 concrete (PCC) (Li et al., 2017a, 2017c). Nitrogen and phosphorus from fertilizers are even more easily lost in PCC  
63 based on the theory of ‘fertilizer moves with water’ (Li et al., 2010; Alberts and Moldenhauer, 1981), indicating  
64 that fertilizers should be continually supplied to meet the nutritional demands of plant growth, which is  
65 labor-consuming and cost-ineffective. More serious is the high alkalinity (PH=13.5) of PCC due to the formation of  
66 C-S-H gel and calcium hydroxide (Kaminskas et al., 2015; Keren et al., 2015), and the alkali leaching plays a  
67 negative role for plant growth. To our best knowledge, both of disadvantages above, poor fertilizer retention and  
68 alkali leaching, can be mitigated by using sulphoaluminate cement (SAC). Compared with Portland cement, SAC  
69 not only has a lower alkalinity value of 11.5 due to the special hydration products, namely ettringite (AFt,  
70  $C_3A \cdot 3C\$ \cdot H_{32}$ , where C = CaO, A = Al<sub>2</sub>O<sub>3</sub>, \$ = SO<sub>3</sub>, H = H<sub>2</sub>O) and calcium aluminate sulphate hydrate gel (AFm,  
71  $C_3A \cdot C\$ \cdot H_{12}$ ) (Martin et al., 2015; Juenger et al., 2011; Chen et al., 2017), which allows it to create a more  
72 favourable environment for plant growth, but also has the advantages of a lower calcination temperature  
73 (~1250 °C), lower CO<sub>2</sub> emissions and easier grinding with less energy required (Li et al., 2017b; Dong et al., 2014;  
74 Chen et al., 2016; García-Maté and Torre, 2013; Berger and Aouad, 2013), which is more energy efficient and  
75 sustainable. In addition, SAC also allows for reduced construction time because of the fast setting and high early  
76 strength gain (Trauchessec et al., 2015; Liao et al., 2011; Bernardo et al., 2006). More importantly, a number of  
77 previous studies (Li et al., 2017a, 2017c; Huang et al., 2016) demonstrated that the release rate of fertilizer in the  
78 hardened SAC pastes was much slower than that in Portland cement. Therefore, SAC is an excellent candidate to  
79 replace Portland cement for developing PC with improved fertilizer retention.

80 The objective of this paper is to develop a new generation of PC made from SAC as matrix and recycled waste  
81 demolition concrete as coarse aggregates, which is expected to have improved fertilizer retention properties,  
82 reduced the cost and satisfied mechanical performance. Firstly, the influence of different consolidation methods  
83 and mixing order of raw materials on the properties (e.g. the void ratio, water permeability and compressive  
84 strength) of PC was investigated. Secondly, the fertilizer release performance of the developed PC modified by two  
85 kind of fertilizers, urea and diammonium phosphate, was evaluated and compared by measuring the release rate of  
86 nitrogen. Finally, the effect of fertilizers on the alkalinity, drying shrinkage and frost resistance of PC was studied.

## 87 2. Experimental Procedures

### 88 2.1 Materials

89 SAC (Grade 42.5 R according to Chinese National Standards GB20472-2006, manufactured in Special  
90 Cement Co., Ltd of Qufu, China) was used as the binder to fabricate the PC. The chemical composition of the SAC  
91 was analyzed by an X-ray fluorescence spectrometer (XRF, Tiger S8, Germany), as listed in Table 1. X-ray  
92 diffraction (XRD) and quantitative X-ray diffraction (QXRD) were employed to measure and quantify the main  
93 phases of SAC, the results of which are shown in Fig. 2 and Table 2. Crushed limestone rubble and waste  
94 demolition concrete were used as the natural and recycled coarse aggregates, respectively. The physical properties  
95 of both aggregates are represented in Table 3. Fig. 3 shows the lab-based crushing process for the waste demolition  
96 concrete.

97 Urea ( $\text{CO}(\text{NH}_2)_2$ , purity of 99.0 wt. %, Sinopharm Chemical Reagent Co., Ltd, China) and diammonium  
98 phosphate (DP,  $(\text{NH}_4)_2\text{HPO}_4$ , purity of 99.0 wt. %, Sinopharm Chemical Reagent Co., Ltd, China) were used as the  
99 two fertilizers in this study. Boric acid (BA, purity of 99.0 wt. %, Sinopharm Chemical Reagent Co., Ltd, China)  
100 along with polycarboxylate polymer (PP, Shandong Academy of Building Research, China) were incorporated into  
101 the PC as the retarder and superplasticizer, respectively.

## 102 2.2 Experimental

### 103 2.2.1 Sample Preparation

104 Details on the mix proportions and preparation methods of PC are listed in Table 4. SAC was mixed with  
105 double distilled  $\text{H}_2\text{O}$  at the water/cement ( $w/c$ ) ratio of 0.25 (Jang et al., 2015). The weight ratios of  
106 aggregate/cement, BA/cement and PP/cement were fixed at 6.000, 0.004 and 0.011 for all mixes.

107 Three methods (A, B and C) with different mix order were used to fabricate PC in this study, as shown in  
108 Table 5, followed by three different consolidation methods. Fresh mixture was cast into cylindrical molds ( $\phi$ -100  
109 mm  $\times$  200 mm) and cube molds (100  $\times$  100  $\times$  100 mm), and then consolidated by the following three ways: (1)  
110 vibration (V, 3 -15 s), (2) pressure (P, 0.5 - 2.0 MPa) or (3) ramming (R). Specimens were then cured at a  
111 temperature of  $20 \pm 2$  °C and humidity of 95+ 5 % RH. After 24 hours, specimens were demolded and placed back  
112 for further curing under the same conditions until further testing.

### 113 2.2.2 Fertilizer Release Performance

114 Based on the optimum preparation method obtained from the section of 2.2.1, urea and DP were then  
115 individually added to the PC, based on the mix proportions listed in Table 6. The cubic specimens (100  $\times$  100  $\times$  100  
116 mm) were used to measure the compressive strength, fertilizer release rate and alkalinity of pore fluid of the  
117 fertilized PC. Two types of beam specimens (100  $\times$  100  $\times$  515 mm and 100  $\times$  100  $\times$  400 mm) were prepared for  
118 measuring the drying shrinkage rate and frost resistance. Finally, cylinder specimens ( $\phi$ -100  $\times$  200 mm) were

119 prepared for void ratio and water permeability tests. All specimens were cured in the exact same conditions as  
120 described in the section of 2.2.1.

## 121 2.3 Measurements

### 122 2.3.1 Void Ratio Test

123 The total void ratio ( $R_{TV}$ ) was obtained according to the following equation:

$$124 R_{TV} = \left[ 1 - \frac{M_2 - M_1}{\rho_w V} \right] \times 100\%$$

125 where ( $M_1$ ) is the initial mass of the cylinder samples in the water, ( $M_2$ ) is the final mass tested following air drying  
126 ( $25 \pm 1$  °C,  $20 \pm 2\%$  RH) for 24 hours, ( $V$ ) is the sample volume and ( $\rho_w$ ) is the density of water.

127 Similarly, the connected void ratio ( $R_{CV}$ ) was obtained according to the following equation:

$$128 R_{CV} = \left[ 1 - \frac{M_3 - M_1}{\rho_w V} \right] \times 100\%$$

129 where ( $M_3$ ) is the final mass measured following air curing ( $25 \pm 1$  °C,  $95 \pm 2\%$  RH) for 24 hours.

### 130 2.3.2 Coefficient of Water Permeability (CWP)

131 The CWP of planting concrete was measured in accordance with (ISO 17785-1, 2016; Tarnai et al., 2004; Shen  
132 et al., 2013), for which the experimental set-up is shown in Fig. 4. The CWP was determined over a period of 50  
133 seconds under a water head of 150 mm and calculated according to the following equation:

$$134 K_r = \frac{H}{h} \times \frac{Q}{A(t_2 - t_1)}$$

135 where  $K_r$  is the CWP (cm/s),  $H$  is the specimen length (cm),  $Q$  is the volume of water discharge from  $t_1$  to  $t_2$  (cm<sup>3</sup>),  
136  $h$  is the difference of water head,  $t_2 - t_1$  is time (s) and  $A$  is the cross sectional area of the cylindrical specimen (cm<sup>2</sup>).

### 137 2.3.3 Compressive Strength

138 For compressive strength test, three capped cubic specimens were tested at a loading rate of 2 mm/min by a  
139 2000 KN capacity hydraulic testing device (DC-2000, China), according to the Chinese National Standard GB  
140 50081-2002 (GB/T 50081, 2002).

### 141 2.3.4 Fertilizer Release Rate and Alkalinity of Pore Fluid

142 The fertilizer release rate can be divided into two parts: a nitrogen release rate and a phosphorus release rate.  
143 Samples were soaked in a container with 10 L deionized water (Fig. 5) at a temperature of  $20 \pm 1$  °C. At the age of  
144 1 day, 3 days, 7 days and 28 days, the nitrogen and phosphorus concentration of the soaking solution were  
145 measured according to the Chinese National Standards GB 11894-1989 (GB/T 11894, 1989) and GB 11893-1989  
146 (GB/T 11893, 1989), respectively. Correspondingly, pH value of the soaking solution was measured using a  
147 laboratory grade pH meter (pHs-3E, China). It should be noted that after every single test, the container was

148 re-filled with 10 L pure deionized water to avoid the saturation effect.

### 149 2.3.5 Drying Shrinkage

150 The drying shrinkage was determined according to the Chinese National Standard GB 50082-2009 (GB/T  
151 50082, 2009), the set-up of which is shown in Fig. 6. Specimens for the drying shrinkage were cured at  $20 \pm 2$  °C  
152 and 95+% RH for 3 days. After that, testing specimens were cured at a temperature of  $20 \pm 1$  °C and humidity of 65  
153  $\pm 5$  % RH. The length of the specimen was accurately measured at the predetermined times of 1 day, 3 days, 5 days,  
154 7 days, 10 days, 14 days, 20 days, 28 days, 45 days, 60 days, 90 days and 150 days and the drying shrinkage was  
155 calculated based on the following equation:

$$156 \quad \varepsilon = \frac{L_0 - L_t}{L_0}$$

157 where ( $\varepsilon$ ) is the drying shrinkage rate,  $L_0$  is the initial beam length and  $L_t$  is the final beam length.

### 158 2.3.6 Frost Resistance

159 Frost resistance test was carried out according to the Chinese National Standard GB 50082-2009 (GB/T 50082,  
160 2009). The schematic view of frost resistance is shown in Fig. 7. Test samples (100 mm  $\times$  100 mm  $\times$  400 mm)  
161 were prepared according to the fast freeze-thawing method and were cured as standard for 2 days followed by  
162 insertion in  $20 \pm 1$  °C deionized water for 24 hours. The samples were subsequently taken out from deionized water,  
163 surface dried, accurately weighed and recorded as  $m_0$ . A total of 40 freeze-thaw cycles implemented, where one  
164 cycle consisted of freezing at -18 °C for 4 hours and thawing at 18 °C for 2 hours. Samples were then accurately  
165 weighted and recorded as  $m_1$ . The weight loss rate (WLR) of the sample was used as a parameter to reflect the frost  
166 resistance of planting concrete and calculated using the following equation:

$$167 \quad \text{WLR} = \frac{m_0 - m_1}{m_0} \times 100\%$$

## 168 3. Results and Discussion

### 169 3.1 Effect of Vibration Time on the Basic Properties of PC

170 Fig. 8 shows the void ratio and CWP of PC mixed by Method A and consolidated by vibration. In terms of  
171 void ratio, it is clear to see that the total and interconnected void ratios of PC decrease with the increasing vibration  
172 time, as shown in Fig. 8a. For example, the control sample *Aref* (without any consolidation) shows the highest total  
173 and interconnected void ratio at 47.1 % and 43.4 %, which are approximately equal to the void ratio of recycled  
174 aggregate stacking (48.4%, Table 3). It indicates that the stacking of aggregates is not compact if the PC is prepared  
175 without any consolidation. At the longest vibration time of 25 seconds i.e. sample AV5, the total void and  
176 interconnected void ratio of PC sharply decrease to 41.3 % and 36.5 %, respectively. A similar trend can also be

177 found in the CWP of PC, as shown in Fig. 8b. Previous studies (Neithalath et al., 2010; Bhutta et al., 2013)  
178 demonstrated that there was a linear correlation between the void ratio and CWP of the ordinary pervious concrete.  
179 Therefore, the relationship between the interconnected void ratio and CWP of planting concrete can be established,  
180 as shown in Fig. 9. The correlation coefficient between the interconnected void ratio and CWP reaches up to 0.92,  
181 confirming that there is an extremely significant linear correlation between the two parameters similar to the  
182 ordinary pervious concrete. It also suggesting that the CWP of PC made with recycled aggregates depends on its  
183 interconnected void ratio.

184 It is shown in Fig. 10 that the compressive strength of every vibrated PC is higher than that of the  
185 un-vibrated PC. Maximum values at 1 day, 3 days and 28 days reach up to 3.9, 5.1 and 8.5 MPa, respectively, for  
186 the PC with a vibration time of 15 seconds. More importantly, Fig. 11 shows the correlation between the total void  
187 ratio and compressive strength of PC. Fig.11a reveals a strong correlation up to the vibration time of 15 seconds,  
188 while vibration time long than that leads to a decrease in the correlation coefficient between the total void ratio and  
189 compressive strength, as shown in Fig. 11b. This phenomenon can be explained by that the recycled aggregates  
190 which loosely stack before vibration turn more compact with the increasing vibration time. However, with more  
191 than 15 seconds of vibration, the SAC paste is more likely to detach from the aggregates and precipitate to the  
192 bottom of PC mixture, as shown in Fig. 12, leading to a significant reduction of SAC paste as the binder in PC.

### 193 3.2 Effect of Pressure on the Properties of PC

194 Fig. 13 shows the void ratio and CWP of PC mixed by Method A and consolidated by pressure. It evidently  
195 shows that both parameters of PC decrease with the increase of consolidation pressure. When the pressure reached  
196 1.0 MPa i.e. sample AP2, the interconnected and total void ratios of PC decrease by 12.9 % and 10.6 %,  
197 respectively. As shown in Fig. 14, for this consolidation method, the correlation coefficient of interconnected void  
198 ratio and CWP of PC is 0.99, suggesting that there is a strong linear correlation between the two parameters. It  
199 also confirms the conclusion that the water permeability of PC depends on its interconnected void ratio.

200 Fig. 15 presents the effect of consolidating pressure on the compressive strength of PC. The results indicate  
201 that the highest 1-, 3- and 28-day compressive strengths are up to 2.8, 4.9 and 6.3 MPa, respectively, at the  
202 consolidation pressure of 1.0 MPa. Following the correlation relationship between the total void ratio and  
203 compressive strength of PC (consolidated by pressure) (Fig. 16), it can be observed that the correlation coefficient  
204 up to the pressure of 1.0 MPa significantly increases, as shown in Fig. 16a. However, when the consolidation  
205 pressure increases further to 2.0 MPa, the correlation coefficient greatly decreases with the increase of  
206 consolidation pressure (Fig. 16b). The loss of compressive strength and correlation coefficient can be explained by

207 the difference in recycled concrete aggregate composition compared to natural stone aggregates. The recycled  
208 concrete aggregate contains two additional components, namely, adhered mortar and an interfacial transition zone  
209 (ITZ) between the natural aggregate and the original cement mortar (Shi et al., 2016). Therefore, the bonding  
210 between cement paste and recycled aggregate is weaker than between cement paste and natural aggregate (Bhutta  
211 et al., 2013). In addition, ITZ is considered to be the weakest region of the recycled concrete aggregate due to the  
212 existence of numerous cracks and porosity (Zhang et al., 2015). The weakness of the ITZ can also be reflected  
213 from the value of crushing index listed in Table 3, indicating that the crushing index value of recycled concrete  
214 aggregates (18.3%) is twice as much as that of natural aggregates (9.3 %) and the.

### 215 3.3 Effect of Mixing and Consolidating Methods on the Basic Properties of PC

216 Fig. 17 shows the interconnected void ratio and CWP of PC prepared by different mixing and consolidating  
217 methods. For mixing Method A, it is clear to see that the interconnected void ratio and CWP of the concrete  
218 without any consolidation i.e. sample *A ref* (Fig. 8), are the highest among all the mixing methods. In addition,  
219 the interconnected void ratio of the PC consolidated by ramming (sample *AR*) is higher than that of the PC  
220 consolidated by vibrating for 15 seconds (sample *AV3*) and by pressure for 1.0 MPa (sample *AP2*), while the  
221 CWP of sample *AR* is slightly lower than that of sample *AP2*. Furthermore, the 1-, 3- and 28-day compressive  
222 strength of sample *AR* is obviously higher than that of sample *AV3* and sample *AP2*. Therefore, based on the  
223 mixing Method A, PC consolidated by ramming shows the optimum properties. For mixing Method B, the  
224 interconnected void ratio and CWP of PC consolidated by pressure at 1.0 MPa (sample *BP2*) is obviously higher  
225 than that of the concrete consolidated by vibrating for 15 seconds (sample *BV3*) and by ramming (sample *BR*),  
226 and the 1- and 3-day compressive strength of sample *BP2* is similar to that of sample *BV3* and sample *BR*. So,  
227 based on the mixing Method B, PC consolidated by pressure at 1.0 MPa showed the optimum properties. For  
228 mixing Method C, the interconnected void ratio and CWP of PC consolidated by pressure at 1.0 MPa (sample  
229 *CP2*) is obviously higher than that of the PC consolidated by vibrating for 15 seconds (sample *CV3*) and by  
230 ramming (sample *CR*), and the 1-, 3- and 28-day compressive strength of sample *CP2* is similar to that of sample  
231 *CV3* and sample *CR*. Hence, based on the mixing Method C, PC consolidated by pressure at 1.0 MPa shows the  
232 optimum properties. In sample *AR*, sample *BP2* and sample *CP2*, by contrast, the interconnected void ratio of  
233 sample *AR* is similar to that of sample *BP2* and sample *CP2*, and the CWP and 1-, 3- and 28-day compressive  
234 strength of sample *AR* is obviously higher than that of sample *BP2* and sample *CP2*. In summary, mixing  
235 procedure A and consolidation by ramming i.e. sample *AR* show the optimum properties, namely, an  
236 interconnected void ratio of 40.9 %, a CWP of 2.88 cm/s and a 28-day compressive strength of up to 9.5 MPa. In



237 summary, mixing procedure A and consolidation by ramming will be used to prepare PC samples for testing  
238 fertilizer release performance, alkalinity, shrinkage and frost resistance.

### 239 3.4 Fertilizer Release Performance

240 The released amount and cumulative release rate of nitrogen are the two parameters representing the fertilizer  
241 release performance of PC. Fig. 19 shows the release amounts and cumulative release rates of nitrogen of planting  
242 concrete with urea.. It is evident that the cumulative release rate of nitrogen increases for all samples with the  
243 increase of soaking time. Nonetheless for sample *RUI*, the cumulative release rate of nitrogen is only 25.0 % at 28  
244 days, suggesting that the urea is slowly released from the hardened PC if it is added during the mixing process of  
245 raw materials. The decreased release rate of urea partly results from its absorption by aggregates due to the  
246 ‘fertilizer moves with water’ theory (Li et al., 2017a), and the rest of urea is distributed in the hardened SAC paste.  
247 Since the total porosity of the hardened SAC paste can reach 0.09 cm<sup>3</sup>/g (Li et al., 2017b) and the void ratio for PC  
248 is about 40 %, the urea could move from recycled aggregate and hardened paste to water. So, the nitrogen could  
249 dissolve out from hardened PC after the concrete is kept in deionized water. An additional phenomenon worth  
250 noting is that the release amount and cumulative release rate of nitrogen increase with the increase of urea addition  
251 at the same soaking time. For instance, when the urea dosage is 0.44 kg/m<sup>3</sup> (sample *RUI*), 4.4 kg/m<sup>3</sup> (sample *RU2*)  
252 and 7.7 (sample *RU3*) kg/m<sup>3</sup>, the release amounts of nitrogen for 1 day is 7.9 mg/L, 481.6 mg/L and 1017.3 mg/L  
253 and the release amounts of nitrogen for 28 days is 17.7 mg/L, 394.6 mg/L and 571.5 mg/L, respectively. It might  
254 result from that the concentration gradient of nitrogen between inside and outside of the concrete increases with the  
255 increase of urea dosages and the release speed increases with the increase of concentration gradients.

256 Fig. 20 and Fig. 21 show the released amount and cumulative release rate of nitrogen and phosphorus of PC  
257 with DP. The release amounts of nitrogen and phosphorus increases with the increasing amount of DP at the same  
258 soaking time. In addition, the cumulative release rate of nitrogen and phosphorus increase with the increase of  
259 soaking time and the DP addition at same soaking time. However, the 28-day cumulative release rate of nitrogen of  
260 PC with 0.44 kg/m<sup>3</sup> DP (sample *RDI*), 4.4 kg/m<sup>3</sup> DP (sample *RD2*) and 7.7 kg/m<sup>3</sup> DP (sample *RD3*) reaches up to  
261 16.11 %, 45.43 % and 66.75 %, respectively, and the cumulative release rate of phosphorus of sample *RDI*, sample  
262 *RD2* and sample *RD3* for 28 days is only 0.007 %, 0.030 % and 0.033 %, respectively. This phenomenon suggests  
263 that the nitrogen could easily dissolve out from the hardened PC after the concrete is kept in deionized water, while  
264 the phosphorus is difficult to dissolve out. The mechanism behind can be described as below: DP could hydrolyze  
265 to NH<sub>4</sub><sup>+</sup> and HPO<sub>4</sub><sup>2-</sup> ((NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub> → 2NH<sub>4</sub><sup>+</sup> + HPO<sub>4</sub><sup>2-</sup>) and the latter one could further hydrolyze to H<sup>+</sup> and PO<sub>4</sub><sup>3-</sup>  
266 (HPO<sub>4</sub><sup>2-</sup> → H<sup>+</sup> + PO<sub>4</sub><sup>3-</sup>) (Kumar and Behal, 2017; Shah et al., 2016), which can react with divalent cations (Ca<sup>2+</sup>,

267  $Mg^{2+}$ ) to produce insoluble phosphate ( $CaHPO_4$ ,  $Ca_3(PO_4)_2$ ,  $Mg_3HPO_4$  and  $Mg_3(PO_4)_2$ ) (Kokubo et al., 1991;  
268 Kjellin et al., 2016; Castro et al. 2017). Therefore, the cumulative release rate of phosphorus is much lower than  
269 that of nitrogen, and that is the reason why it is difficult for plants to absorb phosphorus from PC with DP and  
270 therefore extra supply of phosphate fertilizer is required for plants growth.

### 271 3.5 Alkalinity

272 Fig. 22 shows the pH value of the pore solution in PC. From Fig. 22a, it can be seen that the alkalinity of PC  
273 prepared with recycled aggregates (sample *Rref*) is lower than that of the concrete prepared by natural aggregate  
274 (sample *N*). This is mainly due to the unit weight of natural aggregates being higher than that of recycled  
275 aggregates under the same weight ratio of aggregate-to-cement (Table 3). So, the amount of SAC used in samples  
276 *N* ( $265 \text{ kg/m}^3$ ) is higher than that used in sample *Rref* ( $220 \text{ kg/m}^3$ ) (Table 6). The result confirms that the recycled  
277 concrete aggregates are suitable in the preparation of PC due to its lower density. The PH value of the pore fluid in  
278 sample *RU1*, *RU2* and *RU3* is 8.29, 8.23 and 8.24 at 28 days, respectively which is similar to that of sample *Rref*.  
279 However, the alkalinity of the pore fluid of PC increases with the increase of DP addition. When the DP dosage  
280 reaches to  $7.7 \text{ kg/m}^3$  (sample *RD3*), the 28-day alkalinity of the pore fluid in sample *RD3* is increased by 14.8 % to  
281 a value of 9.38. The most likely reason is that DP could react with CH to form insoluble phosphate ( $CaHPO_4$  and  
282  $Ca_3(PO_4)_2$ ) and ammonium hydroxide (AH;  $NH_3 \cdot H_2O$ ), which easily transforms into  $NH_4^+$  and  $OH^-$  ( $NH_3 \cdot H_2O$   
283  $\rightarrow NH_4^+ + OH^-$ ), leading to the increased alkalinity. Therefore, it can be concluded that DP is not suitable to prepare  
284 fertilized PC.

### 285 3.6 Drying Shrinkage

286 Fig. 23 displays the drying shrinkage of PC fertilized with Urea and DP. It indicates that the drying shrinkage  
287 of PC increases with the increasing of curing time. When the curing time is up to 150 days, the drying shrinkage  
288 rate of sample *Rref* is  $147.1 \times 10^{-6}$ , which is higher than that of sample *N* (i.e.  $117.7 \times 10^{-6}$ ). Generally, the drying  
289 shrinkage of concrete occurs when the free water stored in the capillary pores evaporates due to a low  
290 relative-humidity environment. This circumstance leads to a humidity gradient which induces the transport of water  
291 particles from the gel to the capillary pores after which it evaporates (Gonzalez-Corominas and Etxeberria, 2016).  
292 Since drying shrinkage can generate internal stress, mass loss and consequently volume reduction of the PC, the  
293 amount of water evaporation from the PC prepared with recycled aggregates is therefore higher than that of the PC  
294 prepared with natural aggregates due to the water absorption of the recycled aggregates being much higher than that  
295 of natural aggregate (Table 3). More importantly, the addition of Urea can effectively reduce the drying shrinkage  
296 rate of PC prepared with recycled aggregates. The experimental results indicate that the dry shrinkage rate of

297 sample *RU2* is only  $117.0 \times 10^{-6}$  at 150 days, which is comparably close to the value obtained from sample *N*. The  
298 reduced drying shrinkage is attributed to the formation of AFm (Huang et al., 2016). With the help of Urea, AFt  
299 generated in the SAC hydration could be transformed to AFm, which shows a better stability in the drying  
300 condition (Li et al., 2017c; Huang et al., 2016). However, DP could increase the drying shrinkage rate of PC  
301 prepared with recycled aggregates. The dry shrinkage rate of sample *RD2* rises to  $176.5 \times 10^{-6}$  at 150 days. Hence,  
302 urea is more suitable than DP in the preparation of PC using recycled aggregates.

### 303 3.7 Frost Resistance Property

304 Fig. 24 shows the frost resistance of PC from a qualitative point of view. The mass loss rate of PC increases  
305 with the increase of freezing and thawing cycles. When the number of cycles reaches to 40, the mass loss rate of  
306 sample *Rref* is 3.77%. At the same time, the mass loss rate of sample *N* is only 2.73%. This result confirms that the  
307 frost resistance of PC prepared by recycled concrete aggregates is worse than the concrete prepared by natural  
308 aggregates. The likely reason is that the inherent frost resistance of recycled aggregates is much lower than that of  
309 natural aggregates (Fig. 24b) due to its high water absorption (Table 3). Therefore, the mass loss rate of the PC  
310 prepared with recycled aggregates is higher than that of the PC prepared with natural aggregates. Furthermore,  
311 addition of urea improves the frost resistance of planting concrete using recycled aggregates while DP has a  
312 counteractive effect. It might result from the fact that urea is beneficial to the hydration and hardening of SAC  
313 (Huang et al., 2016), while DP could react with CH to form insoluble phosphate ( $\text{CaHPO}_4$  and  $\text{Ca}_3(\text{PO}_4)_2$ ) with  
314 unsatisfactory cementitious properties (Steink et al., 1991), which is negative to SAC hardening.

## 315 4. Conclusions

316 Recycled aggregates from demolished concrete (RADC) was adopted to prepare a new generation of planting  
317 concrete (PC) with enhanced fertilizer release performance. The preparation methods, fertilizer release performance,  
318 alkalinity, dry shrinkage and frost resistance of the PC were investigated in the current study. The investigation  
319 supports the following conclusions:

320 (1) RADC was suitable for developing the PC with improved fertilizer release performance. By optimizing the  
321 fabrication procedures (i.e. mixing procedure A: 1) Recycled aggregates and water were mixed together in a  
322 concrete mixer for 30 seconds; 2) SAC was slowly added into the mixture in 60 seconds; 3) The fresh mixture was  
323 further mixed for another 60 seconds and consolidation by ramming) and mix proportions, the interconnected  
324 porosity, water permeability coefficient and 28-day compressive strength of the PC with  $4.4 \text{ kg/m}^3$  addition of  
325 RADC can be increased to 40.9 %, 2.88 cm/s and 6.5 MPa, respectively.

- 326 (2) A novel PC fertilized with urea showed improved nitrogen and release rate. Using a urea addition of 4.4 kg/m<sup>3</sup>,  
327 the cumulative release rate of nitrogen at 28 days of PC reached up to 72.1 %.
- 328 (3) A novel PC modified by RADC showed a stable pore fluid alkaline environment at approximately 8.20 pH,  
329 which was more suitable for plant growing.
- 330 (4) With the addition of urea, the frost resistance of planting concrete can be improved by 12.4\_% with the  
331 reduction of 150-day drying shrinkage rate by 20.5 %. In contrast, the addition of DP resulted in a greater  
332 mass loss rate and was not be recommended for use as a fertilizer in PC.

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483 \_\_\_\_\_

484 **TABLES**

485

486 **Table 1**

487 Chemical compositions of SAC (wt. %)

Oxide	SiO <sub>2</sub>	CaO	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	MgO	K <sub>2</sub> O	Na <sub>2</sub> O	TiO <sub>2</sub>	SO <sub>3</sub>	Los
Result	9.60	45.16	21.64	2.45	1.28	1.38	0.17	1.03	10.73	6.35

488

489 **Table 2**

490 QXRD data of SAC (%)

Mineral	C <sub>4</sub> A <sub>3</sub> S	C <sub>2</sub> S	C <sub>3</sub> S	C <sub>4</sub> AF	C <sub>2</sub> S <sub>H<sub>2</sub></sub>	C <sub>2</sub> S	CaCO <sub>3</sub>
Result	42.7	19.7	13.4	5.36	2.9	2.7	13.26

491

492 **Table 3**

493 Physical properties of crushed concrete aggregates in comparison to commercial aggregates

Aggregate type	Natural aggregate	Recycled aggregate
Gradation (mm)	20-25	20-25
Crush index (%)	9.3	18.3
Density (kg/m <sup>3</sup> )	2730	2590
Water absorption (%)	0.6	5.6
Water absorption for 5 min (%)	0.6	3.3
Unit weight (kg/m <sup>3</sup> )	1590	1320
Absolute volume (%)	58.5	51.6

494

495

496

497 **Table 4**

498 Mix proportion and preparation method of planting concrete

Sample Name	Unit weight (kg/m <sup>3</sup> )					Mixing procedure	Consolidating method	Time/s	Pressure/MPa
	SAC	Aggregate	water	BA	PP				
Aref	220	1320	55	0.87	2.39	A	-	-	-
AV1							V	5	-
AV2								10	-
AV3								15	-
AV4								20	-
AV5								25	-
AP1							P	-	0.5
AP2								-	1.0
AP3								-	2.0
AR							R	-	-
BV3						B	V	a	-
BP2							P	-	b
BR							R	-	-
CV3						C	V	a	-
CP2							P	-	b
CR							R	-	-

499 a: 15 seconds optimum vibrating time; b: 1.0 MPa optimum consolidation pressure.

500

501 **Table 5**

502 Raw materials mixing methods for preparing planting concrete

Method A	Method B	Method C
1) Recycled aggregates and water were mixed together in a concrete mixer for 30 seconds;	1) Recycled aggregates were soaked in water until the weight maintained no more than 0.5 %;	1) Recycled aggregates and half of water were mixed for 30 seconds in a concrete mixer.
2) SAC was slowly added into the mixture in 60 seconds;	2) Kept the recycled aggregates in a saturated-surface dry (SSD) condition;	2) Half of SAC was added and mixed for another 60 seconds.
3) The fresh mixture was further mixed for another 60 seconds..	3) SSD recycled aggregates and half of SAC were mixed in a concrete mixer for 30 seconds;	3) Left water and SAC were added into the mixture above with further mixing of 120 seconds.
	4) The left SAC and water were added into the mixture above with	

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further mixing of 120 seconds.

---

503

504 **Table 6**

505 Mix proportions of planting concrete with fertilizer (kg/m<sup>3</sup>)

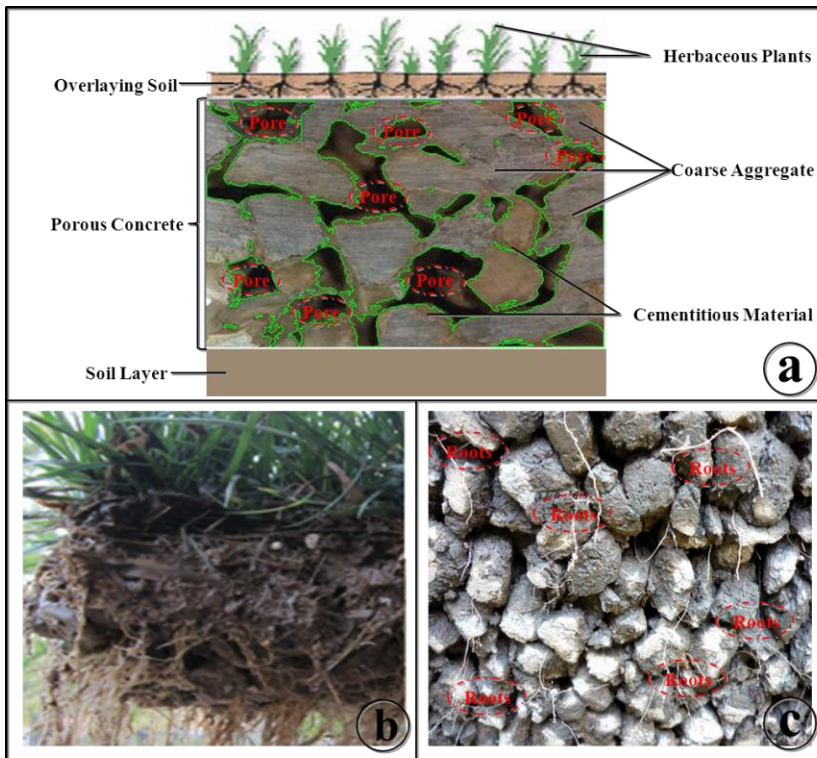
Sample Name	Aggregate type		SAC	Water	Urea	DP	BA	PP
	Recycled	Natural						
N	-	1590	265	66	-	-	1.06	2.92
Rref	1320	-	220	55	-	-	0.87	2.39
RU1		-			0.44	-		
RU2		-			4.4	-		
RU3		-			7.7	-		
RD1		-			-	0.44		
RD2		-			-	4.4		
RD3		-			-	7.7		

506

507

508 **FIGURES**

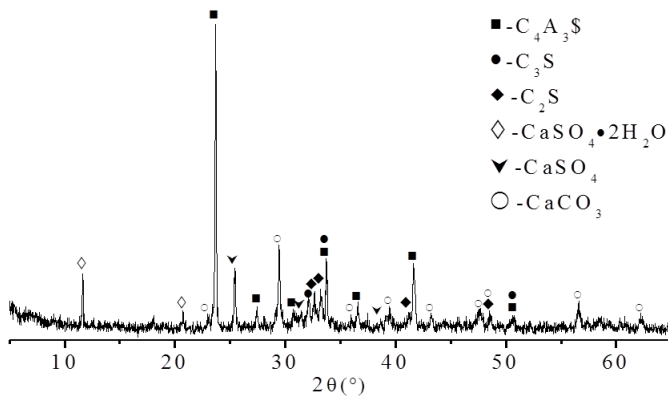
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510

511 Fig.1 (a) Sketch map of planting concrete; (b) Actual-effect pictures of planting concrete; (c) Actual- effect pictures of  
512 plants' roots through the concrete.

513

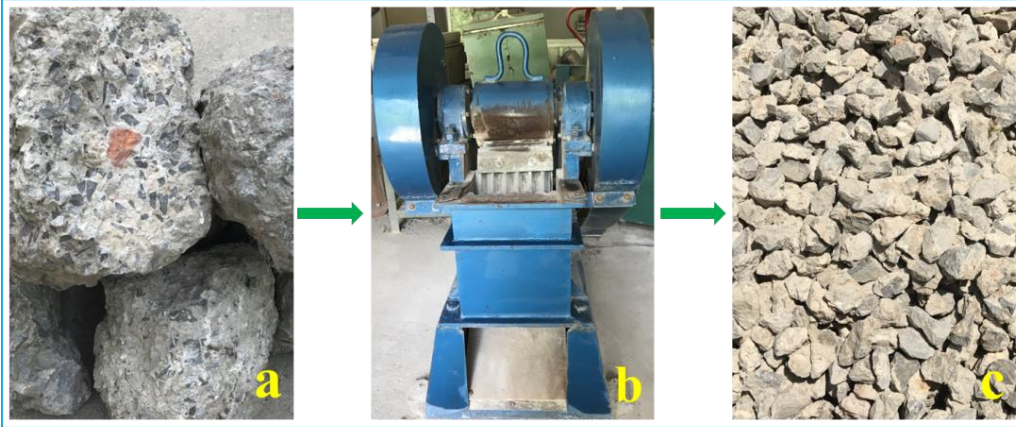


514

515 Fig. 2 X-ray analysis result of SAC.

516

517



518

519 Fig. 3 The preparation process of recycled aggregates showing (a) the waste demolition concrete (b) the jaw crusher

520 (c) the final crushed demolition concrete as aggregates

521

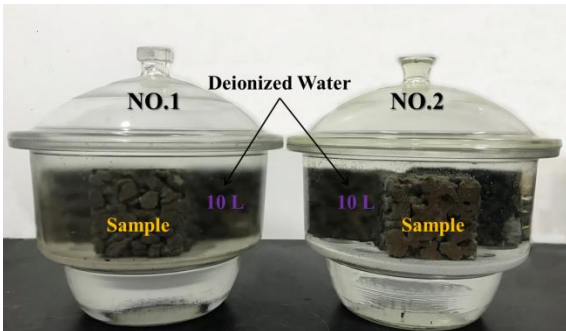


522

523 Fig. 4 CWP set-up for planting concrete.

524

525

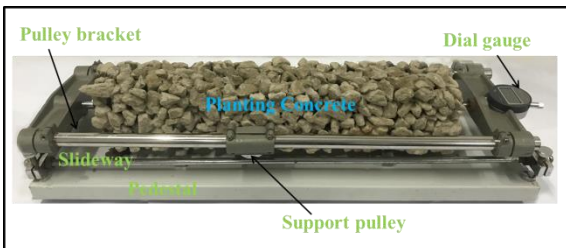


526

527 Fig. 5 Pictures of the test of fertilizer release rate and alkalinity of pore fluid.

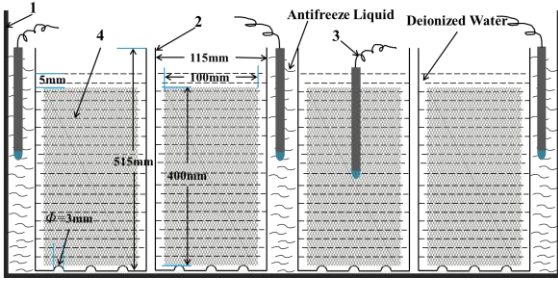
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531 Fig. 6 Drying shrinkage test set-up of planting concrete.

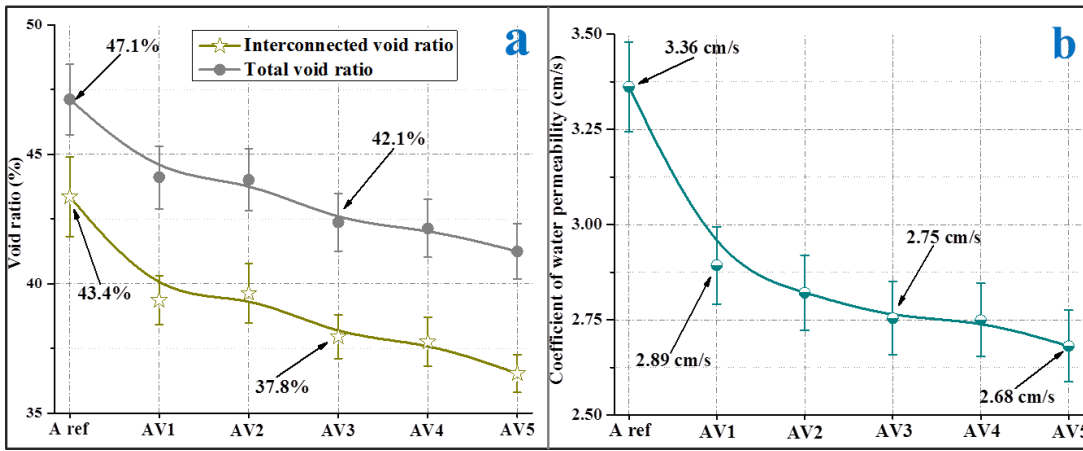


532

533 Fig.7 Schematic view of frost resistance test. 1, Freeze-thawing cycle machine; 2, Rubber container (115 mm × 115  
534 mm × 515 mm); 3, Temperature Sensor; 4, Sample (100 mm × 100 mm × 400 mm).

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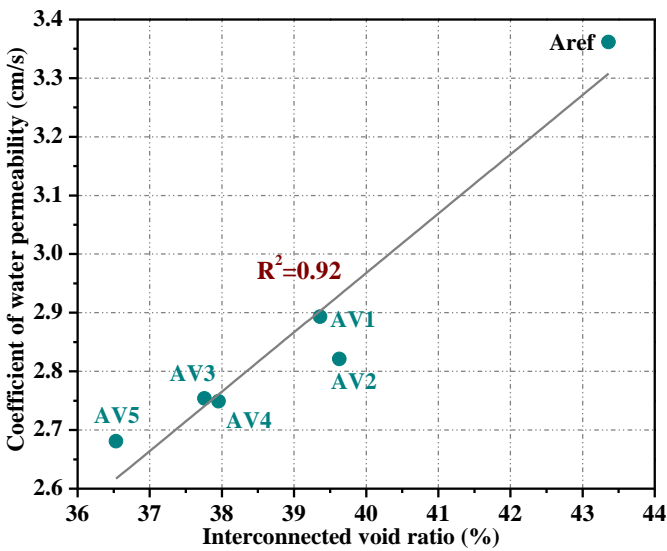
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538 Fig. 8 (a) Void ratio and (b) CWP of planting concrete (consolidated by vibration).

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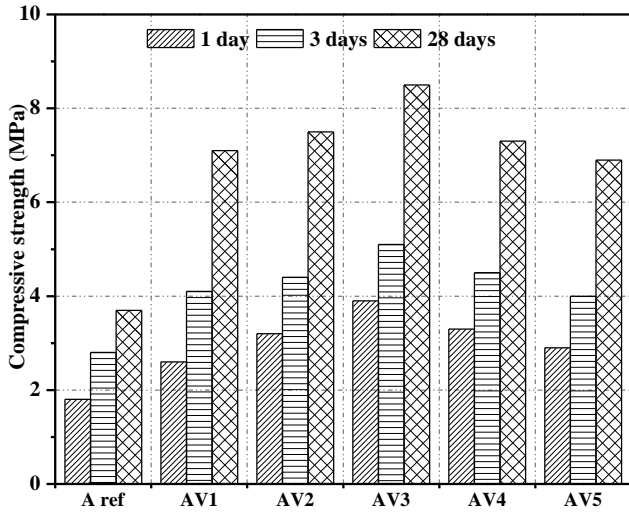


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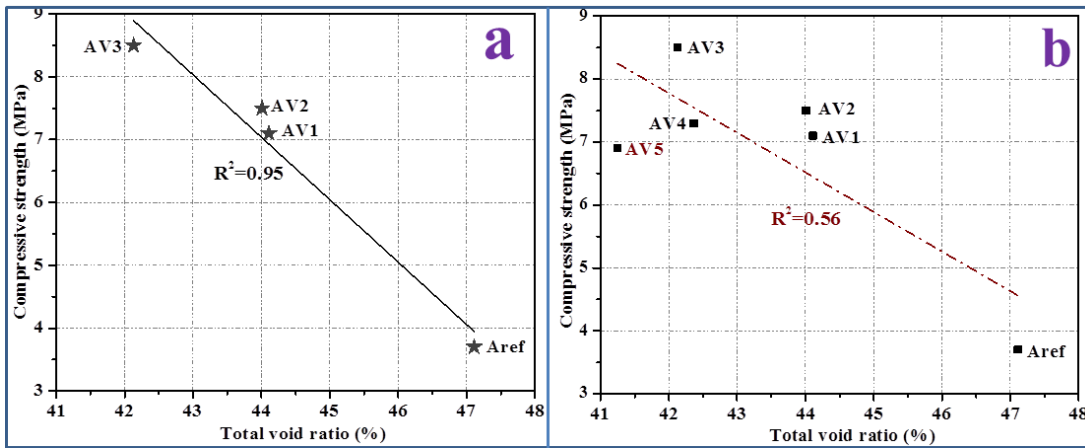
541 Fig. 9 Relationship between interconnected void ratio and CWP of planting concrete (consolidated by vibration).

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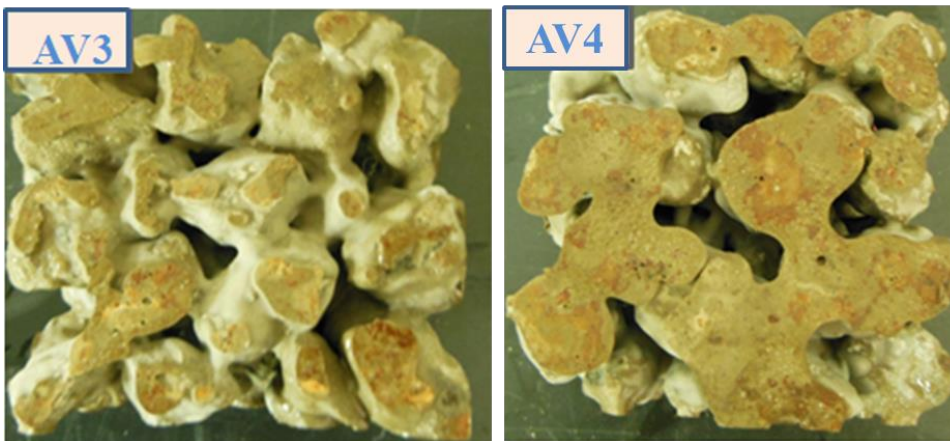




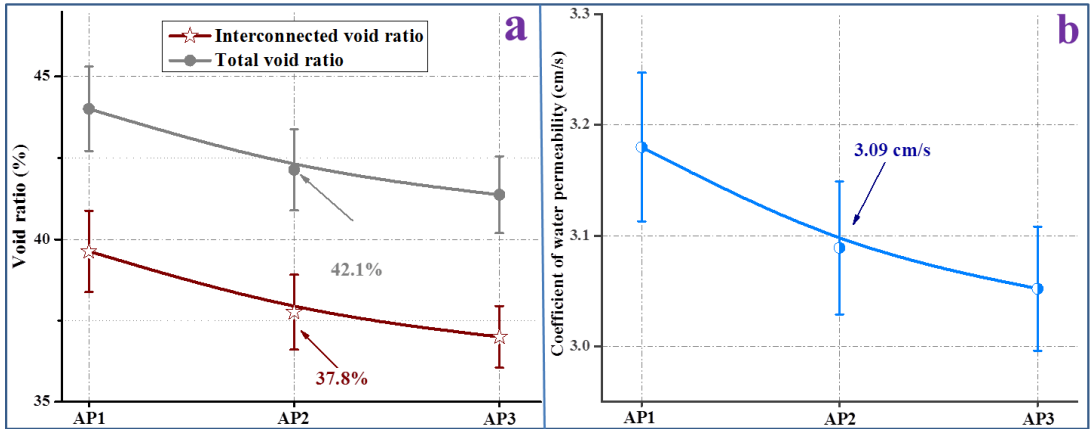
543  
544 Fig. 10 Compressive strength of planting concrete (consolidated by vibration).  
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546  
547 Fig. 11 Relationship between total void ratio and 28 days compressive strength of planting concrete (consolidated  
548 by vibration).  
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550  
551 Fig. 12 Images of the bottom of (a) sample AV3 and (b) AV4 consolidated by vibration.  
552

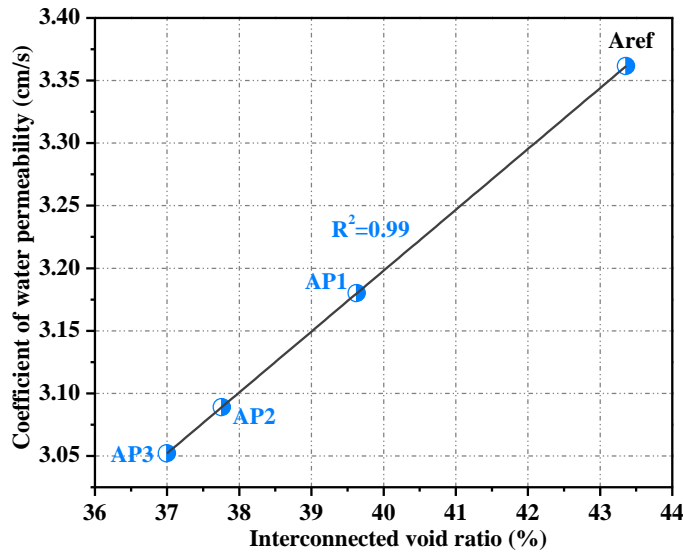


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Fig. 13 (a) Total void and interconnected void ratio and (b) CWP of planting concrete (consolidated by pressure).

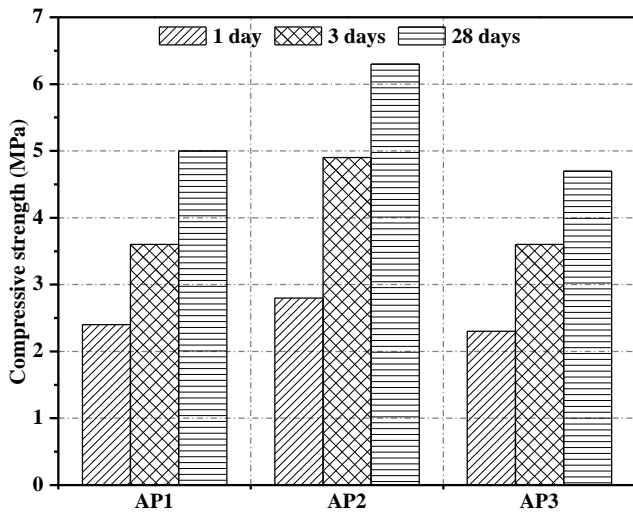


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Fig. 14 Relationship between interconnected void ratio and CWP of planting concrete (consolidated by pressure).

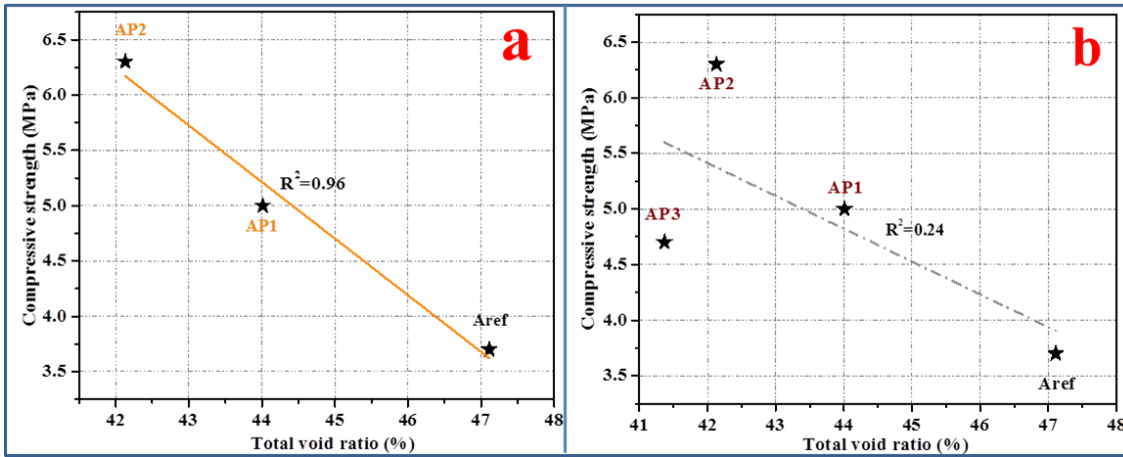


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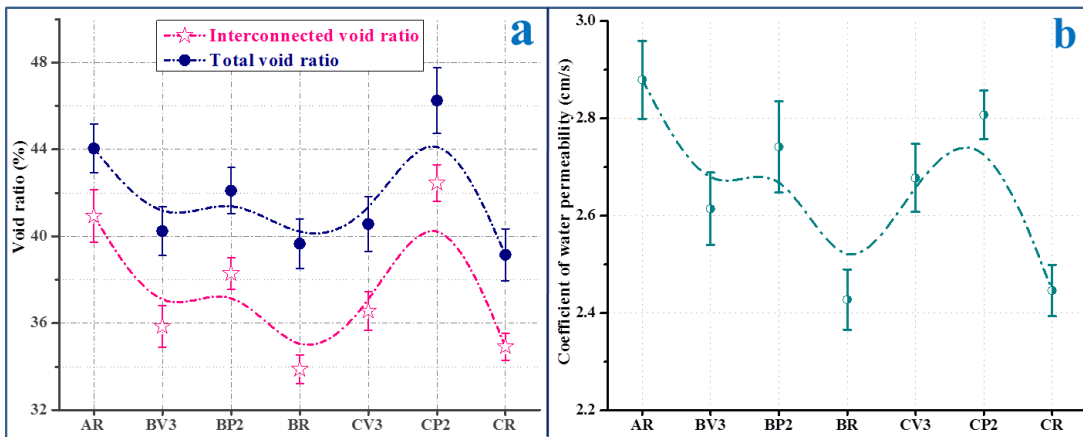
Fig. 15 Compressive strength of planting concrete (consolidated by pressure).



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563 Fig. 16 Relationship between total void ratio and 28-day compressive strength of planting concrete (consolidated  
564 by pressure).

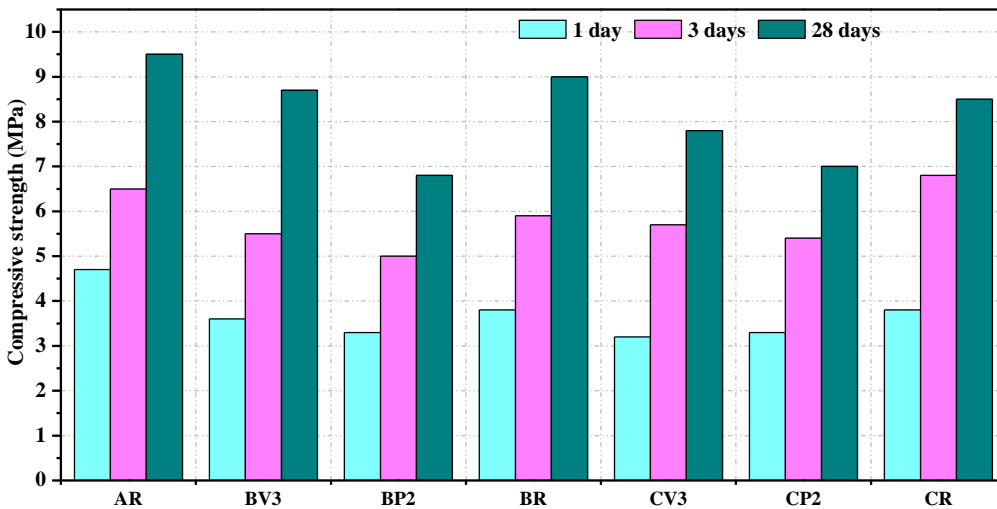
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567 Fig. 17 (a) Total void and interconnected void ratio and (b) CWP of planting concrete.

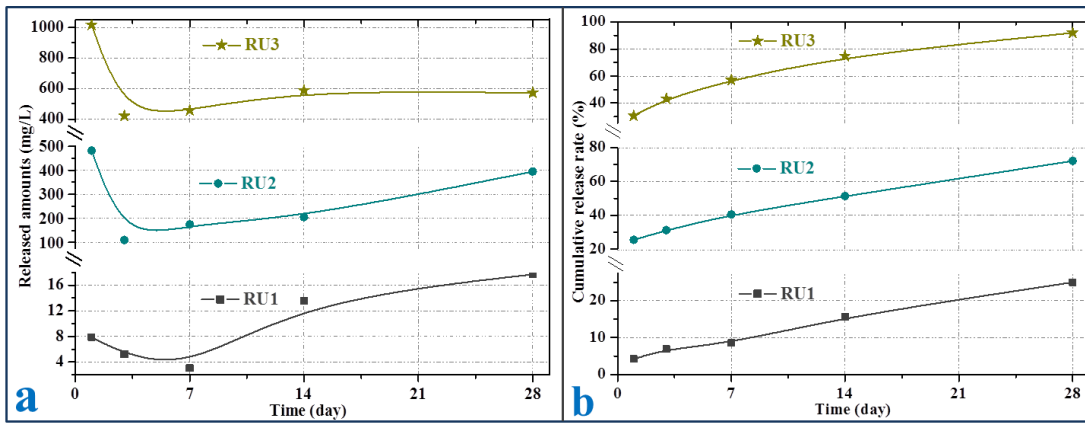
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570 Fig. 18 Compressive strength of planting concrete.

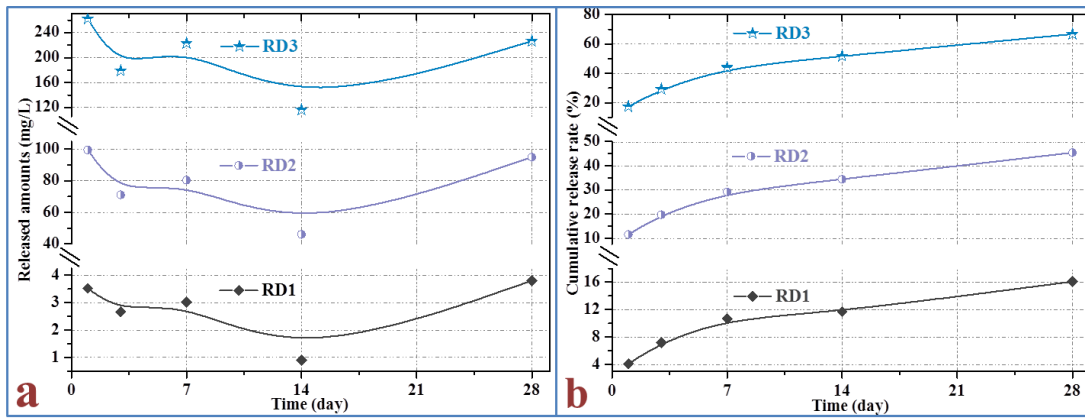
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573 Fig. 19 Release amounts and cumulative release rates of nitrogen of planting concrete with urea.

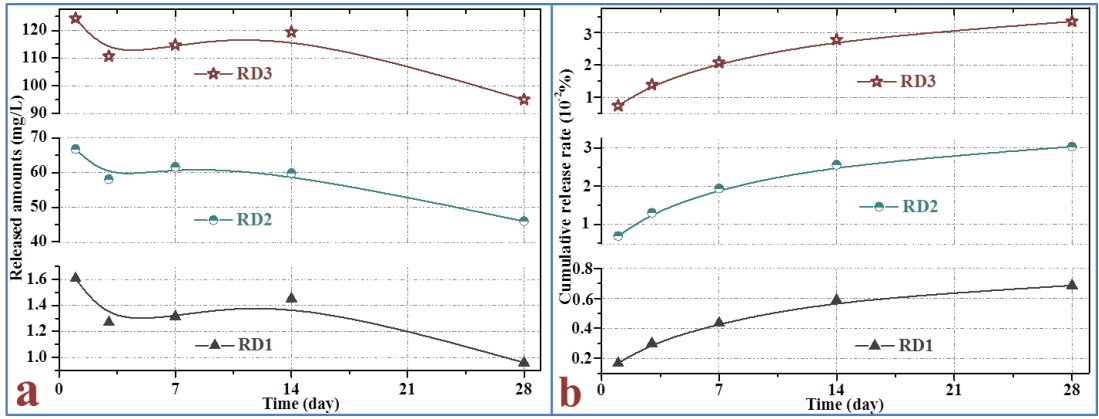
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576 Fig. 20 Release amounts and cumulative release rate of nitrogen of planting concrete contains DP.

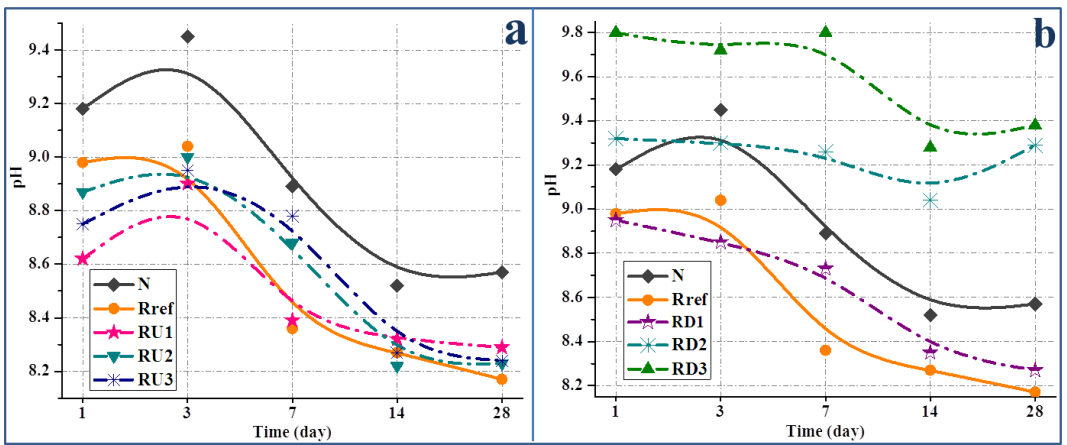
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579 Fig. 21 Release amounts and cumulative release rate of phosphorus of planting concrete contains DP.

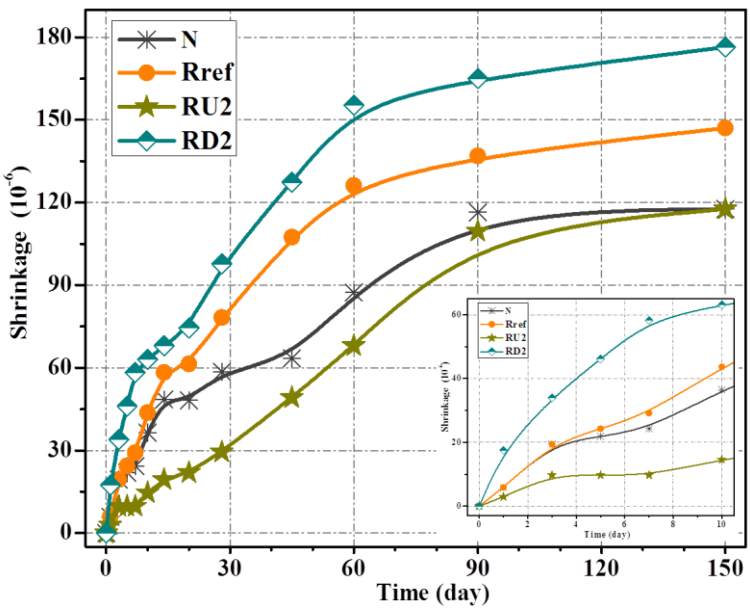
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582 Fig. 22 Alkalinity of pore fluid of planting concrete at various ages.

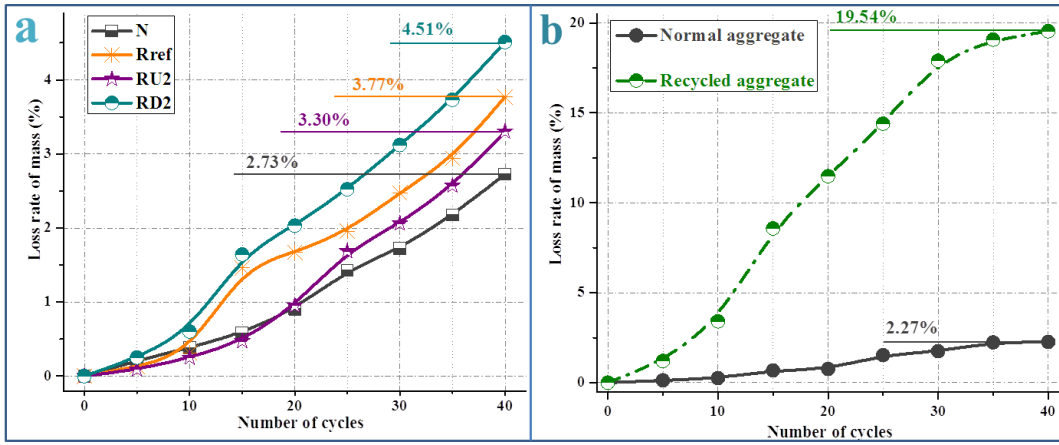
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585 Fig. 23 Dry shrinkage of planting concrete at various ages.

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Fig. 24 Frost resistance (i.e. in terms of mass loss) of planting concrete.

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